

Experiment 35**Advance Study Assignments: Spot Tests for Some Common Anions**

1. Each of the observations in the following list was made on a different solution. Given the observations, state which ion studied in this experiment is present. If the test is not definitive, indicate that with a question mark.
 - A. Addition of 6 M NaOH and Al to the solution produces a vapor that turns red litmus blue.
Ion present: _____
 - B. Addition of 6 M HCl produces a vapor with an acid odor.
Ion present: _____
 - C. Addition of 6 M HCl produces an effervescent.
Ion present: _____
 - D. Addition of 6 M HNO₃, plus 0.1 M AgNO₃, produces a precipitate.
Ion present: _____
 - E. Addition of 6 M HNO₃, plus 1 M BaCl₂, produces a precipitate.
Ion present: _____
 - F. Addition of 6 M HNO₃, plus 0.5 M (NH₄)₂MoO₄, produces a precipitate.
Ion present: _____
2. An unknown containing one or more of the ions studied in this experiment has the following properties:
 - A. No effect is observed on addition of 6 M HNO₃.
 - B. No effect is observed on addition of 0.1 M AgNO₃ to the solution from Part A.
 - C. A white precipitate is observed on addition of 1 M BaCl₂ to the solution from Part A.
 - D. A yellow precipitate is observed on addition of (NH₄)₂MoO₄ to the solution from Part A.

On the basis of this information, which ions are present, which are absent, and which are in doubt?

Present

Absent

In doubt

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M Walker



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Table of Contents Spot Tests Advance Study Assignment

1. Understanding the eBook Spot Tests Advance Study Assignment
 - The Rise of Digital Reading Spot Tests Advance Study Assignment
 - Advantages of eBooks Over Traditional Books
2. Identifying Spot Tests Advance Study Assignment
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Spot Tests Advance Study Assignment
 - User-Friendly Interface
4. Exploring eBook Recommendations from Spot Tests Advance Study Assignment
 - Personalized Recommendations
 - Spot Tests Advance Study Assignment User Reviews and Ratings
 - Spot Tests Advance Study Assignment and Bestseller Lists
5. Accessing Spot Tests Advance Study Assignment Free and Paid eBooks
 - Spot Tests Advance Study Assignment Public Domain eBooks
 - Spot Tests Advance Study Assignment eBook Subscription Services
 - Spot Tests Advance Study Assignment Budget-Friendly Options
6. Navigating Spot Tests Advance Study Assignment eBook Formats

- ePub, PDF, MOBI, and More
- Spot Tests Advance Study Assignment Compatibility with Devices
- Spot Tests Advance Study Assignment Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Spot Tests Advance Study Assignment
 - Highlighting and Note-Taking Spot Tests Advance Study Assignment
 - Interactive Elements Spot Tests Advance Study Assignment
- 8. Staying Engaged with Spot Tests Advance Study Assignment
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Spot Tests Advance Study Assignment
- 9. Balancing eBooks and Physical Books Spot Tests Advance Study Assignment
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Spot Tests Advance Study Assignment
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Spot Tests Advance Study Assignment
 - Setting Reading Goals Spot Tests Advance Study Assignment
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Spot Tests Advance Study Assignment
 - Fact-Checking eBook Content of Spot Tests Advance Study Assignment
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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