

Unit 10



UNIT 10: PERIODIC TABLE: CONCEPTS AND ACTIVITY

Section 10.1: Introduction

In your notebook, read about the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electronic Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 10. It exists in the periodic table.
- _____ 10. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 20. Elements that are strong oxidising agents.
- _____ 20. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 20. There are two representative elements of elements.
- _____ 20. There are two periods of the elements in the periodic table.
- _____ 20. Group 17 elements except for hydrogen and fluorine are the alkali metals.
- _____ 20. Group 17 elements are the alkali metal metals.
- _____ 20. Group 17 elements are highly reactive elements known as halogens.
- _____ 20. Group 17 elements are very reactive elements known as representative elements.
- _____ 20. Alkali metals have properties of both metals and non-metals.

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M Lipman



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