

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	18
_____	Hydrogen
_____	Li (103)
_____	Neon (1018)

Atomic Mass	19.0
Atomic Number	18.0
Electron Configuration	1S ²
Chemical Name	Ne
Chemical Symbol	Ne

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has no the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word as shown in **Model 10.1**.

- _____ 24. There are two representative elements of elements.
- _____ 25. There are two periods of the elements in the periodic table.
- _____ 26. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 27. Group 1A elements are the alkali metals.
- _____ 28. Group 1A elements are highly reactive elements in the periodic table.
- _____ 29. Group 1A elements are very reactive elements found in the periodic table.
- _____ 30. Alkali metals have properties of both metals and nonmetals.

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David Baud



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