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Stoichiometry

Study Guide: Stoichiometry and Yield Calculations

Stoichiometry is the branch of chemistry that deals with the quantitative relationships between reactants and products in chemical reactions. It's essentially about figuring out how much of each ingredient you need to make a specific amount of "product" in a chemical reaction. To do this, we use balanced chemical equations, which show the mole ratios of reactants and products. These mole ratios are crucial for performing stoichiometric calculations.

Mote Ratios of Chemical Equations

- 1.Definition: Mole ratios are the proportions of reactants and products in a chemical reaction, represented by the coefficients in a balanced chemical equation.
- Usage: These ratios allow for the conversion between amounts of different substances in a reaction.
- 3.Example: For the reaction $2H_2*O_2*2H_2O$, the mole ratio of H_2 to O_2 is 2:1, and the mole ratio of H_2 to H_2O is 1:1.

Stolomiometry

- 1.Definition: Stoichiometry involves calculating the quantities of reactants and products in chemical reactions based on the balanced equation.
- 2.Steps:
 - 1. Write and balance the chemical equation.
 - 2. Convert quantities of known substances into moles.
 - Use mole ratios to calculate the moles of the desired substance.
 - 4. Convert moles back into desired units (grams, liters, etc.).

Predicting Amounts of Reactants with Stoichiometry Calculations

- 1.Identify: Determine the amount of one reactant given and find the amount of another reactant or product needed.
- 2.Example: For the reaction N₂*3H₂ * 2NH₃, if you have 5 motes of N₂, you need 5*3*15 motes of H₂ to fully react.

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RJ Shavelson

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web october 2003 a prolific painter draftsman and etcher rembrandt van rijn is usually regarded as the greatest artist of holland s golden age he worked first in his native leiden and from 1632 onward in amsterdam where he had studied briefly ca 1624 with the influential history painter pieter lastman

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web from approximately 1620 to 1624 25 rembrandt trained as an artist as was quite common in his time he had two masters in succession rembrandt's first master was the leiden painter jacob van swanenburgh 1571 1638 with whom according to orlers he remained for about three years van swanenburgh must have taught him the basic skills

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web an innovative and prolific master in three media he is generally considered one of the greatest visual artists in the history of art and the most important in dutch art history unlike most dutch masters of the 17th century rembrandt s works depict a wide range of style and subject matter from portraits and self portraits to landscapes

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web we know that at least two of rembrandt s pupils went into independent art careers govaert flinck and ferdinand bol their art reflects to some degree rembrandt s influence which is expected as they learned to paint in his style however with time their style developed into their own list of rembrandt s pupils and collaborators