

**Section 2.1 – continued**

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
18
Hydrogen
1.008
Group 1

Atomic Mass	19.
Atomic Number	18.
Electron Configuration	17.
Chemical Name	16.
Chemical Symbol	15.

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It exists in the periodic table.
- _____ 19. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing ability.
- _____ 22. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in bold; change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements are located in the periodic table are metals.
- _____ 25. Group 1A elements except for hydrogen are located in the alkali metals.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 6A elements are very reactive elements known as oxygen family.
- _____ 29. Alkaloids have properties of both metals and non-metals.

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M. B. Bergen



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