Name _____ Date _____ Class _____

CHAPTER 1 STUDY GUIDE FOR CONTENT MASTERY

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your notebook, review what you learned about ozone layer.

Use each of the terms below just once to complete the passage.

atmosphere	oxygen gas	ozone	ozone hole
stratosphere	troposphere	ultraviolet radiation	

Earth is **(1)** _____ is made up of several layers. The air we breathe makes up the lowest level. This layer is called the **(2)** _____. The next layer up is called the **(3)** _____. This level contains a protective **(4)** _____ layer.

Chlorine atoms from **(5)** _____ is made by ultraviolet radiation in the upper part of the atmosphere. The ozone forms a layer around Earth, which absorbs **(6)** _____. Without ozone, you are more likely to get a sunburn or possibly skin cancer. The thinning of the ozone layer, called the **(7)** _____ is worldwide because without ozone all organisms on Earth are subject to more harm from more radiation.

In your notebook, review what you learned.

For each statement below, write true or false.

_____ **8.** CFC is another name for a chlorofluorocarbon.

_____ **9.** CFCs are made up of carbon, fluorine, and oxygen.

_____ **10.** All CFCs are synthetic chemicals.

_____ **11.** CFCs usually react readily with other chemicals.

_____ **12.** CFCs were developed as replacements for toxic refrigerants.

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