

**Section 2.1 – continued**

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
18
Hydrogen
1.008
Group 1

Atomic Mass	19.
Atomic Number	18.
Electron Configuration	17.
Chemical Name	16.
Chemical Symbol	15.

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It exists in the periodic table.
- _____ 19. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing ability.
- _____ 22. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements have lines in the periodic table are metals.
- _____ 25. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 29. Alkaloids have properties of both metals and non-metals.

Study Guide For Content Mastery Answers Chapter 5

Joan Ferrante-Wallace



Study Guide For Content Mastery Answers Chapter 5:

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other helpful materials for you to use as you prepare for the exam

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