

Name: _____ Date: _____

MASS TO MOLE STOICHIOMETRY

Answer the questions. Write answers on new paper.

1. In photosynthesis, water reacts with carbon dioxide to give oxygen and glucose ($\text{C}_6\text{H}_{12}\text{O}_6$). Write and balance the chemical equation. How many moles of CO_2 are required to make 120.0g of glucose?



a. What is the mass of potassium nitrate that is produced when 2.04 moles of potassium phosphate react?

b. If 580g of aluminum phosphate are formed, how many moles of aluminum nitrate reacted?



a. If 150g of N_2O_4 was produced, how many moles of O_2 were required?

b. If 4.0×10^3 moles of oxygen reacted, how many grams of N_2 were needed?



a. If you have 5.50 mol of CaC_2 , how much C_2H_2 do you get?

b. How many moles of water are needed when 65.0g of CaC_2 have reacted?

Study Guide For The Mole Answers Chemistry

James E. Brady, John R. Holum



Study Guide For The Mole Answers Chemistry:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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