

UNIT 1 LONG QUIZ

General Direction: Read carefully the instructions before answering.

½ Length-wise

TEST 1

Direction: Write the answer that corresponds to the definition.

1. It occupies space and has definite mass.
2. A classification of matter that mainly consist of Carbon atoms.
3. The three phases of matter are solid, gas and _____.
4. The Law of matter that states that in a given type of chemical substance, the elements are always combined in the same proportions by mass.
5. What is the approximate diameter of an atom?
6. Who formulated the Law of Conservation of Mass?
7. A type of change of matter that does not alters the chemical composition of a matter.
8. It is a compound that chemically named as NaCl (Sodium Chloride).
9. The Law of matter that states that the weights of one element that combine with a fixed weight of another of these elements another.
10. An English Chemist who developed the theory that all elements (matter) are composed of atom.

TEST 2

Directions: If the statement implies truth, write **TRUE; if not, then write **FALSE**. Again, read the statement carefully.**

Based on Dalton's Theory

1. Atoms can be created or destroyed.
2. Atoms of the different element are similar in shape and mass, but differ from the atoms of other elements.
3. Atom is the smallest unit of matter that can take part in a chemical reaction.
4. Atoms of different elements may combine with each other in a fixed, simple, whole number ratios to form compound atoms.
5. All matter consists of indivisible particles called atoms.

TEST 3

Direction: Group the following processes accordingly. (PHYSICAL CHANGE and CHEMICAL CHANGE)

1. Absorption of oil.
2. Oxidation/ Rusting of Iron tools.
3. Decomposition of leaves.
4. Cutting a wood.
5. Combustion of LPG.

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M Carnoy



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