Name:

Date:

#### Student Exploration: Unit Conversions 2 – Scientific Notation and Significant Digits

[Note to teachers and students: This lesson is designed to be a follow-up to the Unit Conversions Student Exploration sheet. The same Gizmo is used for both activities.]

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: resolution, scientific notation, significant digits

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.) Philip measures a room using his feet. (His feet are each about a foot long.) He estimates the room measures about ten and a half feet by thirteen and a half feet. He calculates the room's area to be:  $10.5' \times 13.5' = 141.75$  ft<sup>2</sup>.

- Which do you think is the best description of the area of the room? Highlight your choice.
  - A. The room's area is exactly 141.75 ft<sup>2</sup>. B. The room's area is about 142 ft<sup>2</sup>.
- 2. Explain your choice: I chose that answer because it the answer but rounded up to 3

#### Gizmo Warm-up

When scientists report a value, they have to take several things into consideration. Sometimes values are very large or small. In this case, scientists can use shorthand called scientific notation to report the value.

Scientists also must consider how precise their value is. The precision of a measurement can be shown by the number of significant digits in the value.

To begin, check that the Burj Khalifa question is shown. Drag the three tiles shown below to determine the tower's height in micrometers. The answer is given in scientific notation.

0.828 Richesters - 1000 Pressure - 1000 Pressure - 1000 Pressure - 1000 Pressure - 8.28 - 10<sup>8</sup> micrometers - 8.28 - 10<sup>8</sup> micrometers

- The Burj Khalifa is 828,000,000 micrometers tall. How is this value written in scientific notation in Gizmo<sup>™</sup>? 8.2 \* 10<sup>8</sup>
   the
- 2. In a value, any non-zero digit is considered a significant digit. (Zeroes may or may not be What is the minimum number of significant digits in 828 significant.)

# <u>Unit Conversions 2 Scientific Notation And Significant</u> <u>Digits Answers</u>

**Donald A. McQuarrie, Stanley Gill** 

#### **Unit Conversions 2 Scientific Notation And Significant Digits Answers:**

Chemistry: 1001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H. Langley, 2022-06-08 Practice your way to a better grade in your Chemistry class Chemistry 1001 Practice Problems For Dummies gives you 1 001 opportunities to practice solving problems on all the topics covered in your chemistry class in the book and online Get extra practice with tricky subjects solidify what you ve already learned and get in depth walk throughs for every problem with this useful book These practice problems and detailed answer explanations will catalyze the reactions in your brain no matter what your skill level Thanks to Dummies you have a resource to help you put key concepts into practice Work through multiple choice practice problems on all Chemistry topics covered in class Step through detailed solutions to build your understanding Access practice questions online to study anywhere any time Improve your grade and up your study game with practice practice practice The material presented in Chemistry 1001 Practice Problems For Dummies is an excellent resource for students as well as parents and tutors looking to help supplement classroom instruction Chemistry 1001 Practice Problems For Dummies 9781119883531 was previously published as 1 001 Chemistry Practice Problems For Dummies 9781118549322 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product **Study Guide to Accompany** Buckwalter/Riban College Physics Marllin Lee Simon, 1987 Pharmacy Calculations for Pharmacy Technicians Bradley J. Wojcik, PharmD, 2018-01-15 Are you a pharmacy technician or pharmacy technician student who wants to learn a few simple methods of solving pharmacy calculations without a bunch of formulas Would you like to raise your hand in Pharmacy Calculations Class after the instructor explains a complicated formula and ask to approach the white board to show the class a much simpler method Do you want to go out on your externship and teach practicing pharmacy technicians how to preform pharmacy calculations Do you want to walk into your Pharmacy Calculations Class on the first day knowing that you can ace all the tests before the course begins If you answered yes to any of these questions this book is for you The book s first chapter covers the following auxiliary subjects which are important to a well rounded knowledge of pharmacy calculations Rounding Numbers Roman Numerals The Metric System Scientific Notation Significant Figures Percent Error The Apothecary Avoirdupois Household Systems The second chapter will teach you that all the following types of calculations can be performed with one simple method If you can convert 5 q to mq using this method you can solve the most complicated IV flow rate problem Unit Conversions Dosage Calculations IV Flow Rate Calculations Percent Calculations Percent Strength Calculations Ratio Strength Calculations Quantity to Dispense Calculations Milliequivalent Calculations The third chapter covers concentrations and dilutions While there is not one method of solving all these problems you will quickly see that they all have common components Topics covered are Preparing a Solution Using Two Different Strength Solutions Preparing a Solution from a Stock Solution and a Diluent Calculating the Percent Strength of a Mixture Powder Volume Calculations

Serial Dilution The book includes plenty of exercises to hone your skills along with a self assessment exercise Finally the book ends with a couple of Pharmacy Calculation Puzzles These puzzles are for those students who want to say to themselves If I can solve these I can solve any possible problem I will encounter Water Resources George M. Hornberger, Debra Perrone, 2019-09-03 A scientifically rigorous text grounded in socioeconomic reality that examines both physical hydrology and contemporary water usage issues The fair allocation and wise use of fresh water presents significant challenges across the world To avoid unresolvable crises in the future judiciously managing water resources in the twenty first century is fundamentally important Integrating the underlying science of hydrology with real world usage scenarios Water Resources offers a nuanced modern treatment of contemporary water resource management issues In this ground breaking new text renowned environmental scientist and educator George M Hornberger and award winning environmental engineer Debra Perrone examine the role of water resources in natural social and human built systems helping students understand and evaluate the complex tradeoffs required to achieve sustainable water management Providing a much needed educational tool that looks at freshwater resources within the context of the crucial water energy food nexus the text includes a primer on the elements of physical hydrology necessary to understand resource availability covers rivers lakes groundwater and soil water relates water to agriculture energy urbanization and the environment highlights connections between water quantity and quality explains the economic and legal constraints around water resources considers the impacts of climate change and population growth and proposes paths forward for the sustainable use of water Teaching basic methods used to make informed water management decisions the book includes illustrative quantitative calculations qualitative think pieces and case studies An appendix provides a review of units dimensions and conversions useful for addressing each chapter s example problems Online answer keys are also available Positioned to become the foremost text on water resource issues this companion to Hornberger's widely regarded Elements of Physical Hydrology reveals the enormity of the water crisis facing the planet while offering realistic hope CHEMICAL PROCESS CALCULATIONS PRASAD, RAM, 2022-04-13 The present textbook is written for undergraduate students of chemical engineering as per the syllabus framed by AICTE curriculum It explains the basic chemical process principles in a lucid manner SI units chemical stoichiometry and measures of composition behaviour of gases vapour pressure of pure substances and humidity and saturation are covered in detail In addition mass and energy balances of chemical processes have also been described Chemical processes without chemical reactions include fluid flow mixing evaporation distillation absorption and stripping liquid liquid extraction leaching and washing adsorption drying crystallization and membrane separation process SALIENT FEATURES Description of all concepts and principles with a rich pedagogy for easy understanding Correct use of SI units Over 270 solved examples for understanding the basic concepts Answers to all chapter end numerical problems for checking the accuracy of calculations TARGET AUDIENCE BE B Tech Chemical Engineering *Introduction to General, Organic, and Biochemistry Morris* 

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courses in water engineering and also serves as a valuable resource to professionals operating and managing water and wastewater treatment plants The chapters in this book are standalone offering the flexibility to choose combinations of topics to suit the requirements of a given course or professional application Features Contains example problems and diagrams throughout to illustrate and clarify important topics Problems both in SI and USC system of units The procedure of unit cancellation followed in all solutions to the problems Design applications and operation of water and wastewater system emphasized Includes numerous practice problems with answers and discussion questions in each chapter cover a range of engineering interventions to help conserve water resources and preserve water quality Elements of Environmental Chemistry Jonathan D. Raff, Ronald A. Hites, 2020-07-09 A practical approach to environmental chemistry Elements of Environmental Chemistry 3rd Edition provides readers with the fundamentals of environmental chemistry and a toolbox for putting them into practice This is a concise accessible and hands on volume designed for students and professionals working in the chemical and environmental sciences The 3rd Edition has been completely revised and rearranged The first chapter on tool skills has been expanded to include thermodynamic considerations and measurement issues The former chapter on the partitioning of organic compounds has been expanded to cover the fates of organic compounds with an emphasis on developing the readers chemical intuition for predicting a chemicals fate based on structure The material on lead mercury pesticides PCBs dioxins and flame retardants has been expanded and combined into the last chapter and supplemented with more references to the literature The problem sets have been extended and now include over 130 problems some of which An Inquiry into Science Education, Where the Rubber Meets the Road Richard N. can be solved using Excel Steinberg, 2012-01-01 An inquiry into science education is an exploration into education in a context that is grounded and significant It is written by a college professor of Physics and Science Education who spent sabbatical year as a full time science teacher in a neighborhood high school in a poor area of New York City His varied experiences highlight the contrast of what science education is and what it can be The framework through which the book is written is that science education should be an active purposeful process which promotes functional understanding and critical thinking Science learners should be given the opportunity to build an understanding of benchmark principals of science based on their own observations and reasoning In much the same way this book explores benchmark principals of science education through real classroom experiences Standard approaches of teaching and assessment are presented and alternative opportunities are described Theories and strategies of science education emerge from analysis of classroom observations Although the focus is on the teaching and learning of science the subtext is implications of a failing educational system and what can be done about it The primary intended audience is educators of all capacities but particularly science teachers An inquiry into science education integrates critical topics of science education in a contextualized accessible and easy to read narrative The secondary intended audience is non fiction readers This book examines educational issues relevant to a general audience

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