Chemistry Unit 8 Worksheet 4 Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

 Calculate the number of moles of potassium chlorate, KClO₃ (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	2 KClO ₃ +	≥ 2 KCl +	3 02
Before:	1.2		0
Change:	-1.2	+1.2	+1.8
Afters	.0	1.2	1.8

1.2 moles KCRO, needed

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equations	Non-	2 HCl >	MgCl ₂ +	882
Before:	0.100	7		0
Change:	-0.100	-0.200 +	0.100 +	0.100
Afters		2	0.100	0.100

0.200 protes HCI needed

0.73 g x 1 mole = .0040 mole 183 gr

 Ethane, C₂H₆ reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	2 C2H6	+ 7 O ₂	→ 4 CO₂	+ 6 H ₂ O
Before:	2.20	5.5	0	0.0
Change:	-2.20	~ 7.70	+ 4,40	+ 6,60
After:	0	3.5	4.40	6.60

O2 needed: 7,70 mole x 32.0 g = 246 g O2 required I mode

 Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation: Ni(NO3)2 + 2 NaOH --> Ni(OH)2 + 2 NaNO3

> Ni(NO₃)₂ + 2 NaOH -> Ni(OH)₂ + 2 NaNO₃ Equation: Before: 0.0040 No. 26 -0.0040- O. GOOSIO **-0.0040** +0.06900

N. S. 0.0080 Afflors 0.0640 NaNO₂ produced: 0.0080 mole x 85.0 g = 0.68 g 1 mode

Change:

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web questions should be asked exactly as stated in the ecdi2030 with no changes to wording except for one question that follows a skip rule as explained later in this manual interviewers must ask all 20 questions the fact that some questions might sound too hard or too easy can generate reactions from respondents

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web explore the resources below to prepare yourself for the competency based interviewing style used in the un system guidance read these tips on preparing for competency based interviews