Solutions Review Worksheet

Names

- L. in a 15% alcohol solution, the solute is ______, and the solvent is ______
- Mixtures characterized by an observable segregation of component substances are called mixtures.
- The process of crystal decomposition into hydrated ions is called
- 4. Solutions that are able to dissolve additional solute are said to be _____; solutions that cannot dissolve any more solute at a given temperature are _____; and solutions that contain more solute than can normally be dissolved at a given temperature are _____.
- 6. Properties that depend only on the number of solute particles are ______ properties.
- Classify each of the following as a pure substance, a heterogeneous mixture or a homogeneous mixture:
 - a. Smoke

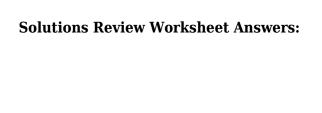
Direct specifies

c. Bubbling alcohol

- 8. A solution is a ______ mixture.
- is an example of a solution of gases.
- 10. Water is an excellent solvent for lonic compounds because each water molecule is
- Write a balanced dissociation reaction for Ca(OH)₁ dissolved in easter.
 (remember to use the appropriate subscripts.)
- £2. What is the difference between a substance like ethanol [CH₂CH₂CH₃CH₃CH₃] dissolving in water and a substance like sodium chloride (NaCl) dissolving in water?
- 13. A 0.50 M solution of NaCl is diluted by taking 50.0 mL of it and adding 150.0 mL of water. What is the diluted concentration of the NaCl solution?
- 14. What mass of CaBr, is contained in 75.0 mL of a 0.50 M solution of CaBr,?

Solutions Review Worksheet Answers

Jicheng Xie



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