

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	21
_____	22
_____	23
_____	24

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It exists in the periodic table.
- _____ 19. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing agents.
- _____ 22. Group 2 elements.

Column B

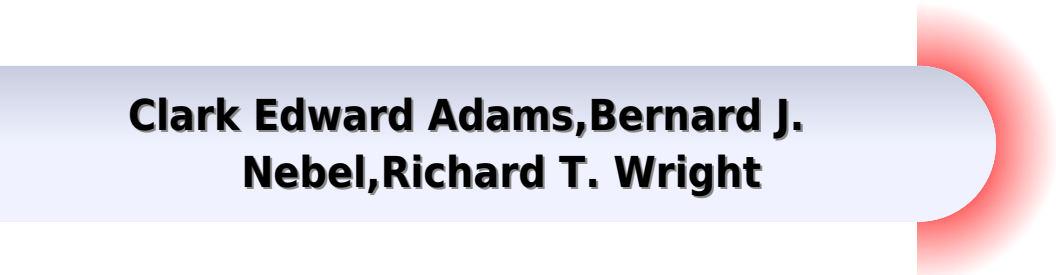
- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in bold; change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements have a line of the periodic table.
- _____ 25. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 26. Group 2 elements are the alkaline earth metals.
- _____ 27. Group 1 elements are highly reactive elements in the periodic table.
- _____ 28. Group 2 elements are very reactive elements in the periodic table.
- _____ 29. Elements have properties of both metals and nonmetals.

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**Clark Edward Adams, Bernard J.
Nebel, Richard T. Wright**



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