

Fire Technology and Arson Investigation

Properties of FIRE:

A. Physical Properties

- 1) **Specific Gravity**- the ratio of the weight of a solid or substance to the weight of an volume of water
- 2) **Vapor density** – the weight of a volume of pure gas compared to the weight of a volume of dry air at the same temperature and pressure.
- 3) **Vapor pressure** – the force exerted by the molecules on the surface of the liquid at equilibrium.
- 4) **Temperature** – the measure of thermal degree of the agitation of molecules of a given substance; the measure of the molecular activity within a substance.
- 5) **Boiling point** – the constant temperature at which the vapor pressure of the liquid is equal to the atmospheric pressure.
- 6) **Ignition temperature or kindling temperature**- the minimum temperature to which the substance the air must be heated in order to initiate or cause self-contained combustion without the addition of heat from outside sources.
- 7) **Fire point**- the temperature at which the material will give off ample vapors to keep burning. There is usually about 5 to 10 degrees differences between the flashpoint and the fire point of most materials. Since these two are just a few degrees apart.
- 8) **Flashpoint**- the temperature at which a material is not hot enough to keep burning, but still gives off enough vapors to cause a flame to "flash" across the surface. The term "flashpoint" is used to express the condition of a fuel vaporizing, whether or not it is vaporizing fast enough to keep burning.

B. Chemical Properties:

1. **Endothermic reaction** – are changes whereby energy (heat) is absorbed or is added before the reaction takes place.
2. **Exothermic reaction** – reactions or changes that release or give off energy (heat) thus they produce substance with less energy than the reactants.
3. **Oxidation** – a chemical change in which combustible material (fuel) and an oxidizing agent react.
4. **Combustion/flame** – the manifestation of fire when the fire is in its gas-phased combustion. A matter that is produced by fire.

Fire - is simply the perceptible phase of burning or combustion, which is the chemical

- **Combination of oxygen in the air with carbon contained in matter.**
- **a form of combustion**

Combustion – is a self-sustaining chemical reaction producing energy or products the cause more reactions of the same kind is an exothermic reaction

Theories of Combustion

Fire Triangle Theory

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