

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 1. A column in the periodic table
- _____ 2. A row in the periodic table
- _____ 3. Group 1 elements
- _____ 4. Elements that are shiny and conduct electricity
- _____ 5. Group 17 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word as shown in **Model 10.1**.

- _____ 1. There are four representative elements of phosphorus.
- _____ 2. About three times fewer of the elements in the periodic table are metalloids.
- _____ 3. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 4. Group 2A elements are the alkaline earth metals.
- _____ 5. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 6. Group 8A elements are very unreactive elements known as noble gases.
- _____ 7. Alkaloids have properties of both metals and more reactive metals.

Study Guide For Content Mastery Chapter 14

Graeme Ching



Study Guide For Content Mastery Chapter 14:

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Urm Abnormal Psychology Durand,Vincent Mark Durand,David H. Barlow,2004-08

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