

Chapter 3 Study Guide

SECTION 3.1 Matter

In your textbook, read about elements and atomic structure.

Use each of the terms below just once to complete the passage.

atom electrons element neutrons nucleus protons

A(n) (1) _____ is a substance that cannot be broken down into simpler substances. A(n) (2) _____ is the smallest particle of matter having all that element's characteristics. It is made up of smaller particles. The (3) _____ is made up of protons and neutrons. Small particles that have mass and positive electrical charges are (4) _____. Particles that have about the same mass as protons, but that are electrically neutral are (5) _____. Surrounding the nucleus of an atom are tiny particles called (6) _____, which have little mass, but have negative electrical charges that are exactly the same magnitude as the positive charges of protons.

In your textbook, read about atomic structure and isotopes.

Complete each statement.

7. The number of protons in an atom's nucleus is the _____.
8. When atoms of the same element have different mass numbers, they are known as _____ of that element.
9. The spontaneous process through which unstable nuclei emit radiation is called _____.
10. A(n) _____ represents the area in an atom where an electron is most likely to be found.
11. An atom that gains or loses an electron and has an electric charge is called a(n) _____.
12. The combined number of protons and neutrons is the _____.
13. The _____ is the average of the mass numbers of the isotopes of an element.

Study Guide Homes Today And Tomorrow Answer

James Elmore Allen



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