

- 2) $\underline{1} \text{ HBr} + \underline{1} \text{ KHCO}_3 \rightarrow \underline{1} \text{ H}_2\text{O} + \underline{1} \text{ KBr} + \underline{1} \text{ CO}_2$
- 3) $\underline{2} \text{ GaBr}_3 + \underline{3} \text{ Na}_2\text{SO}_3 \rightarrow \underline{1} \text{ Ga}_2(\text{SO}_3)_3 + \underline{6} \text{ NaBr}$
- 4) $\underline{3} \text{ SnO} + \underline{2} \text{ NF}_3 \rightarrow \underline{3} \text{ SnF}_2 + \underline{1} \text{ N}_2\text{O}_3$

Using the equation from problem 2 above, answer the following questions:

- 5) If I do this reaction with 30 grams of HBr, how many grams of carbon dioxide will be formed? (2 points)

- 6) Using the following equation:



How many grams of sodium sulfate will be formed if you start with 150 grams of sodium hydroxide and you have an excess of sulfuric acid? (2 points)

- 7) Using the following equation:



How many grams of lithium nitrate will be needed to make 350 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction? (2 points)

Unit 8 Stoichiometry Test Practice

B Lingard

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Unit 8 Stoichiometry Test Practice:

The book delves into Unit 8 Stoichiometry Test Practice. Unit 8 Stoichiometry Test Practice is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Unit 8 Stoichiometry Test Practice, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Unit 8 Stoichiometry Test Practice
 - Chapter 3: Unit 8 Stoichiometry Test Practice in Everyday Life
 - Chapter 4: Unit 8 Stoichiometry Test Practice in Specific Contexts
 - Chapter 5: Conclusion
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