Unit 08 - Stoichiometry - Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?

2KCLO₃ → 2KCl + 3O₂

12 mol KClO₃ 3 mol O₂ 2 mol KClO₃

21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?

2KCLO₃ → 2KCl + 3O₂

14 mol KetO₃ 3 mol O₂ 2 mol KClO₃

__45 mol H₂O___ 3. How many moles of water are produced from 22.5 moles of oxygen?
2H₂ + O₁Z → 2H₂O

22.5 mol-O₂ 2 mol H₂O 1 mol-O₂

_20 mol HCI____ 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

H₂ + Cl₂ → 2HCl

10 mol H₂ 2 mol HCl 1 mol H₂

21 mol Cl₂
5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.

2Fe + 3Cl₂ → 2FeCl₃

14 mol EeCl₃ 3 mol Cl₂ 2 mol FeCl₃

Unit Stoichiometry Problems Answer Key

Basanta Kumara Behera, Ajit Varma

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