Balancing Equations

Constitute All more

Section It

Balance the following chemical equations below. [8]

(a)
$$1 C_1 H_0 O_2 (1) + 5 O_2 (g) \rightarrow 4 H_2 O (g) + 4 CO_2 (g)$$

(b)
$$1 \text{ Al}_2(CO_3)_1(s) \rightarrow 1 \text{ Al}_2O_3(s) + 3 \text{ CO}_2(g)$$

(c)
$$2 \text{ HNO}_2(aq) + 1 \text{ Ni}(s) \rightarrow 1 \text{ Ni}(NO_2)_2(aq) + 1 \text{ Hz}(q)$$

(d)
$$\underline{1}$$
Mn (s) + $\underline{2}$ CuCl (eq) \rightarrow $\underline{2}$ Cu (s) + $\underline{1}$ MnCl₂ (s)

Section 2:

Construct balanced chemical equations for the reactions below. [4]

 (a) Zinc undergoes displacement reaction with aqueous silver nitrate to produce zinc nitrate and silver

 (b) Calcium Carbonate decomposes to give calcium oxide and carbon dioxide

The Learning Odyssey Answer Key Chemistry

Shasha Hu

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