

CHAPTER

STUDY GUIDE

Gases

Lesson 12.1 The Gas Laws

In your textbook, read about the basic concepts of the three gas laws.

Use each of the terms below to complete the passage. Each term may be used more than once.

pressure	temperature	volume
----------	-------------	--------

Boyle's law relates (1) _____ pressure _____ and (2) _____ volume _____ if (3) _____ temperature _____ and amount of gas are held constant. Charles's law relates (4) _____ temperature _____ and (5) _____ volume _____ if (6) _____ pressure _____ and amount of gas are held constant. Gay-Lussac's law relates (7) _____ pressure _____ and (8) _____ temperature _____ if (9) _____ volume _____ and amount of gas are held constant.

In your textbook, read about the effects of changing conditions on a sample of gas.

For each question below, write *increases*, *decreases*, or *stays the same*.

- | | |
|----------------------|--|
| Increases _____ | 10. The room temperature increases from 20°C to 24°C. What happens to the pressure inside a cylinder of oxygen contained in the room? |
| Stays the same _____ | 11. What happens to the pressure of the gas in an inflated expandable balloon if the temperature is increased? |
| Decreases _____ | 12. An aerosol can of air freshener is sprayed into a room. What happens to the pressure of the gas if its temperature stays constant? |
| Increase _____ | 13. The volume of air in human lungs increases before it is exhaled. What happens to the temperature of the air in the lungs to cause this change, assuming pressure stays constant? |
| Decrease _____ | 14. A leftover hamburger patty is sealed in a plastic bag and placed in the refrigerator. What happens to the volume of the air in the bag? |
| Increase _____ | 15. What happens to the pressure of a gas in a lightbulb a few minutes after the light is turned on? |

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IM Harris



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Table of Contents Study Guide Gases

1. Understanding the eBook Study Guide Gases
 - The Rise of Digital Reading Study Guide Gases
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Gases
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Gases
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Gases
 - Personalized Recommendations
 - Study Guide Gases User Reviews and Ratings
 - Study Guide Gases and Bestseller Lists
5. Accessing Study Guide Gases Free and Paid eBooks
 - Study Guide Gases Public Domain eBooks
 - Study Guide Gases eBook Subscription Services
 - Study Guide Gases Budget-Friendly Options

6. Navigating Study Guide Gases eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide Gases Compatibility with Devices
 - Study Guide Gases Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide Gases
 - Highlighting and Note-Taking Study Guide Gases
 - Interactive Elements Study Guide Gases
8. Staying Engaged with Study Guide Gases
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Gases
9. Balancing eBooks and Physical Books Study Guide Gases
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Gases
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide Gases
 - Setting Reading Goals Study Guide Gases
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide Gases
 - Fact-Checking eBook Content of Study Guide Gases
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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