

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	_____
_____	_____
_____	_____
_____	_____
_____	_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ **16.** It appears in the periodic table.
- _____ **17.** It has no line periodic table.
- _____ **18.** Group 1 elements.
- _____ **19.** Elements that are shiny and conduct electricity.
- _____ **20.** Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ **21.** There are two representative elements of elements.
- _____ **22.** Most elements have lines in the periodic table for metalloids.
- _____ **23.** Group 1A elements except for hydrogen are metals in the solid state.
- _____ **24.** Group 2A elements are the alkaline earth metals.
- _____ **25.** Group 7A elements are highly reactive nonmetals known as halogens.
- _____ **26.** Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ **27.** Elements have properties of both metals and nonmetals.

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RS Peters



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