

Chemistry VIVA Questions & Answers

1) Explain the term Aliquot and Diluent?

- Aliquot : It is a measured sub-volume of original sample
- Diluent: Material with which sample is diluted

2) Explain what is molality?

Molality is the number of solute that is present in 1 kg of a solvent.

3) Explain what is titration?

Titration is a process to determine the molarity of a base or an acid. In this process a reaction is carried out between the known volumes of a solution with a known concentration, against the known volume of a solution with an unknown concentration.

4) Explain what is buffer?

A buffer is an aqueous solution which has highly stable pH. It is a blend of a weak acid and its conjugate base or vice versa. On adding small amount of base or acid to buffer, its pH hardly changes.

5) Explain how buffer works?

In buffer when hydrogen ion is added, it will be neutralized by the base in buffer. Hydroxide ion will be neutralized by the acid. On the overall pH of the buffer solution, these neutralization reactions will not show much effect.

Viva Chemistry Question Of Chapter Abbottabad

Er. Meera Goyal



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Practical Chemistry S. K. Agarwal, S. K. Wadhwa, 1982 **Viva Catchup: Chemistry For Class X** Nivedita Bose, 2009-01-01 **Viva Catchup: Chemistry** Mitch Fry, 2007-01-01 **Viva Studymates: Chemistry** Aleyamma Ninan, 2010-01-01 *Practical/Laboratory Manual Chemistry Class - XI* Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod

2 To bend the glass rod at an angle

3 To draw a glass jet from a glass tube

4 To bore a cork and fit a glass tube into it

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2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique

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2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method

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3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample

4 To prepare the pure crystals of copper sulphate from the given crude sample

5 To prepare pure crystals of benzoic acid from a given impure sample

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3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper

2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration

3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper

4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH

5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

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4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions

2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions

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5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method

2 To prepare M 10 solution of sodium carbonate by direct weighing method

3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid

4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

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6 Qualitative Analysis

Analysis of Anions

Analysis of Cations

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7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test

2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

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INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions

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2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation

2 To purify water by boiling

3 To purify water by distillation method

4 To purify water by reverse osmosis technique

5 To purify water by GAC method

6 To purify water by bleach treatment

7 To purify water by oxidising agent

8 To purify water by ozone treatment method

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3 Water Analysis

1 To test the hardness of different water samples

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4 Foaming Capacity of Various Soaps

1 To compare the foaming capacity of different washing soaps

2 To study the effect of addition of sodium

carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

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Table of Contents Viva Chemistry Question Of Chapter Abbottabad

1. Understanding the eBook Viva Chemistry Question Of Chapter Abbottabad
 - The Rise of Digital Reading Viva Chemistry Question Of Chapter Abbottabad
 - Advantages of eBooks Over Traditional Books
2. Identifying Viva Chemistry Question Of Chapter Abbottabad
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Viva Chemistry Question Of Chapter Abbottabad
 - User-Friendly Interface
4. Exploring eBook Recommendations from Viva Chemistry Question Of Chapter Abbottabad
 - Personalized Recommendations
 - Viva Chemistry Question Of Chapter Abbottabad User Reviews and Ratings
 - Viva Chemistry Question Of Chapter Abbottabad and Bestseller Lists
5. Accessing Viva Chemistry Question Of Chapter Abbottabad Free and Paid eBooks
 - Viva Chemistry Question Of Chapter Abbottabad Public Domain eBooks
 - Viva Chemistry Question Of Chapter Abbottabad eBook Subscription Services
 - Viva Chemistry Question Of Chapter Abbottabad Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Viva Chemistry Question Of Chapter Abbottabad Compatibility with Devices
 - Viva Chemistry Question Of Chapter Abbottabad Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Viva Chemistry Question Of Chapter Abbottabad
 - Highlighting and Note-Taking Viva Chemistry Question Of Chapter Abbottabad
 - Interactive Elements Viva Chemistry Question Of Chapter Abbottabad
8. Staying Engaged with Viva Chemistry Question Of Chapter Abbottabad
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Viva Chemistry Question Of Chapter Abbottabad
9. Balancing eBooks and Physical Books Viva Chemistry Question Of Chapter Abbottabad
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Viva Chemistry Question Of Chapter Abbottabad
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Viva Chemistry Question Of Chapter Abbottabad
 - Setting Reading Goals Viva Chemistry Question Of Chapter Abbottabad
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Viva Chemistry Question Of Chapter Abbottabad
 - Fact-Checking eBook Content of Viva Chemistry Question Of Chapter Abbottabad
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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