

Temperature Mix

Purpose

To predict the final temperature of a mixture of cups of water at different temperatures

Apparatus

3 Styrofoam cups
liter container with a wide mouth
thermometer (Celsius)
pail of cold water
pail of hot water

Discussion

If you mix a pail of cold water with a pail of hot water, the final temperature of the mixture will be between the two initial temperatures. What information would you need to predict the final temperature? You'll begin with the simplest case of mixing **equal** masses of hot and cold water.



Procedure

Step 1: Begin by marking your three Styrofoam cups equally at about the three-quarter mark. You can do this by pouring water from one container to the next and mark the levels along the inside of each cup.

Step 2: Fill the first cup to the mark with hot water from the pail, and fill the second cup with cold water to the same level. Measure and record the temperature of both cups of water.

Temperature of cold water = 23 °C

Temperature of warm water = 39 °C

Step 3: Predict the temperature of the water when the two cups are combined. Then pour the two cups of water into the liter container, stir the mixture slightly, and record its temperature.

Predicted temperature = 31 °C

Actual temperature of water = 31 °C



If there was a difference between your prediction and your observation, what may have caused it?

Pour the mixture into the sink or waste pail. (Don't be a klutz and pour it back into either of the pails of cold or hot water!) Now you'll investigate what happens when **unequal** amounts of hot and cold water are combined.

Temperature Mix Lab Manual Physics

William Suddards Franklin



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Table of Contents Temperature Mix Lab Manual Physics

1. Understanding the eBook Temperature Mix Lab Manual Physics
 - The Rise of Digital Reading Temperature Mix Lab Manual Physics
 - Advantages of eBooks Over Traditional Books
2. Identifying Temperature Mix Lab Manual Physics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Temperature Mix Lab Manual Physics
 - User-Friendly Interface
4. Exploring eBook Recommendations from Temperature Mix Lab Manual Physics
 - Personalized Recommendations
 - Temperature Mix Lab Manual Physics User Reviews and Ratings
 - Temperature Mix Lab Manual Physics and Bestseller Lists
5. Accessing Temperature Mix Lab Manual Physics Free and Paid eBooks
 - Temperature Mix Lab Manual Physics Public Domain eBooks
 - Temperature Mix Lab Manual Physics eBook Subscription Services
 - Temperature Mix Lab Manual Physics Budget-Friendly Options

6. Navigating Temperature Mix Lab Manual Physics eBook Formats
 - ePub, PDF, MOBI, and More
 - Temperature Mix Lab Manual Physics Compatibility with Devices
 - Temperature Mix Lab Manual Physics Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Temperature Mix Lab Manual Physics
 - Highlighting and Note-Taking Temperature Mix Lab Manual Physics
 - Interactive Elements Temperature Mix Lab Manual Physics
8. Staying Engaged with Temperature Mix Lab Manual Physics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Temperature Mix Lab Manual Physics
9. Balancing eBooks and Physical Books Temperature Mix Lab Manual Physics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Temperature Mix Lab Manual Physics
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Temperature Mix Lab Manual Physics
 - Setting Reading Goals Temperature Mix Lab Manual Physics
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Temperature Mix Lab Manual Physics
 - Fact-Checking eBook Content of Temperature Mix Lab Manual Physics
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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