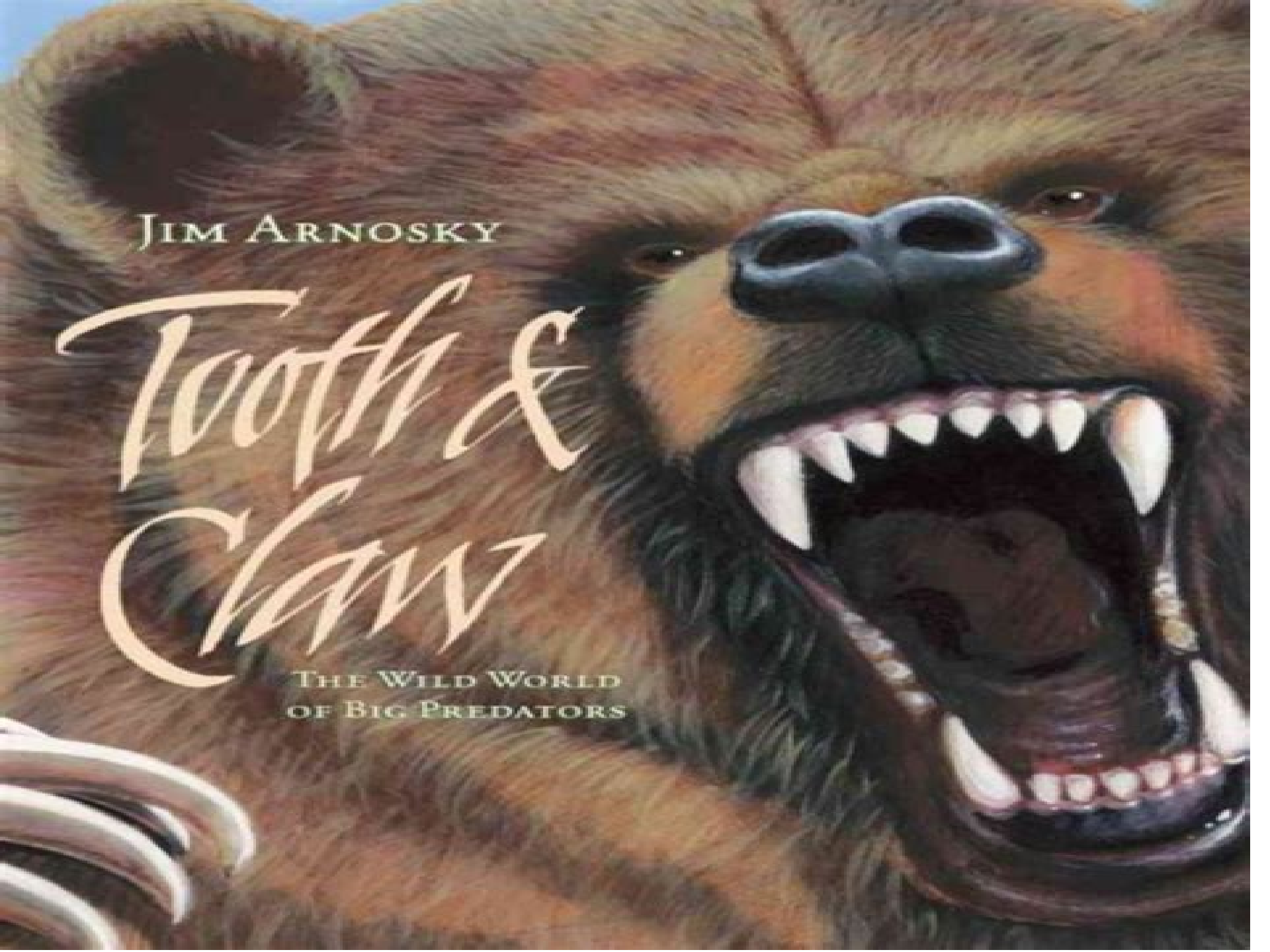




JIM ARNOSKY

Tooth & Claw

THE WILD WORLD
OF BIG PREDATORS

Tooth And Claw

Léo Pomar



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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$
 $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$
 $2\text{Pb} + \text{O}_2 \rightarrow 2\text{PbO}$
 $2\text{HCl} + \text{Ca}(\text{OH})_2 \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$
 $2\text{Al} + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3$
 $2\text{KBr} + \text{Cl}_2 \rightarrow 2\text{KCl} + \text{Br}_2$
 $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$
 $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$
 $2\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 $2\text{Fe} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2$
 $2\text{Fe} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2$
 $2\text{Fe} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2$

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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$
 $2\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow 3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr} + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{GaSO}_3 + 3\text{H}_2\text{NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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