

combining aqueous solutions of  $\text{CoCl}_2$  and  $\text{AgNO}_3$ .

- A)  $\text{Co}^{2+}$  and  $\text{NO}_3^-$     B)  $\text{NO}_3^-$  and  $\text{Cl}^-$     C)  $\text{Co}^{2+}$  and  $\text{Ag}^+$     D)  $\text{Cl}^-$     E)  $\text{NO}_3^-$

14. The solubility of potassium nitrate in water at  $35^\circ\text{C}$  is about 60 grams  $\text{KNO}_3$  per 100 grams of water. How many grams of  $\text{KNO}_3$  should dissolve in 300 grams of water at  $35^\circ\text{C}$ ?

- A) 180 grams    B) 20 grams    C) 335 grams    D) 220 grams

15. Select the most acidic compound from the following.

- A)  $\text{SO}_3$     B)  $\text{Al}_2\text{O}_3$     C)  $\text{CaO}$     D)  $\text{PbO}$     E)  $\text{H}_2\text{O}$

16. Which of the following is totally dissociated when mixed with water?

- A)  $\text{HC}_2\text{H}_3\text{O}_2$     B)  $\text{HF}$     C)  $\text{H}_2\text{SO}_4$     D)  $\text{HNO}_3$     E)  $\text{HBr}$

17. Which solution below has the highest concentration of hydroxide ions?

- A)  $\text{pH} = 3.21$     B)  $\text{pH} = 12.6$     C)  $\text{pH} = 7.93$     D)  $\text{pH} = 9.82$     E)  $\text{pH} = 7.00$

18. According to the Arrhenius theory, what does a base do?

- A) accepts  $\text{H}^+$  ions from an acid  
B) ionizes completely when dissolved in water  
C) ionizes in aqueous solutions to produce hydronium ions  
D) produces  $\text{OH}^-$  ions when dissolved in water.

19. Apples have a pH of 3.9, beans have a pH of 5.9, and coconuts a pH of 7.9. Which of the following statements is true?

- A) Beans are 100 times more acidic than coconut, and 10 000 times more acidic than apples.  
B) Apples are 100 times more acidic than beans, and 10 000 times more acidic than coconut.  
C) Beans are 100 times more acidic than apples, and 10 000 times more acidic than coconut.  
D) Coconut is 100 times more acidic than beans, and 10 000 times more acidic than apples.

20. A lab procedure involving the careful addition of an acid from a buret is called

- A) titration    B) dilution    C) endpoint    D) standard    E) distillation

### Short Answer

1. Will formic acid,  $\text{CH}_2\text{O}_2$ , be soluble in water? Explain your reasoning. (3C)

# Unit 12 Answers Solubility

**YCT Expert Team**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'YCT Expert Team' text.

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