

Unit 10 Stoichiometry Test 1: Short Answers

Answer all questions including units and balanced equations where appropriate.

1. How many moles of aluminium are produced when completely with 2.5 mol of FeCl_3 ?

$$2\text{Al} + 3\text{FeCl}_3 \rightarrow 3\text{Fe} + \text{Al}_2\text{Cl}_6$$
2. How many grams of FeCl_3 are produced when 2.5 moles of aluminium with oxygen at FeCl_3 ?

$$2\text{Al} + 3\text{FeCl}_3 \rightarrow 3\text{Fe} + \text{Al}_2\text{Cl}_6$$
3. The reaction below shows that decomposition of FeCl_3 is complete. How many grams of oxygen are produced when 25.7g FeCl_3 is heated?

$$2\text{FeCl}_3 \rightarrow 2\text{Fe} + 3\text{Cl}_2$$
4. A reaction that has been calculated to produce 2.5g of FeCl_3 actually produces 2.2g of FeCl_3 . What is the percent yield?
5. Balance the equation and determine the limiting reagent when 25.5g of aluminium react with 25.5g of oxygen according to this equation.

$$4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

6. Limiting Reagent is

7. Amount of Al_2O_3 produced is

8. How actually when 25.5g of FeCl_3 what is the percent yield?

Unit 8 Stoichiometry Test Answers

Kaplan Test Prep



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Solving Problems in Chemistry Rod O'Connor, Charles Mickey, Alton Hassell, 1977 *Neet Chapter-Wise & Topic-Wise Solved Papers: Chemistry (2005-2022) With 5 Mock Test* Dr. K. G. Ojha, 2022-06-02 The knowledge of Chemistry helps you to understand the world around you From food to pharmaceuticals Chemistry plays a huge role in making informed decisions Therefore to brush up your intellect we present the NEET Chapterwise and Topicwise Chemistry Solved Papers 2005 2021 which is designed to provide a simplified yet systematic understanding to ace the examination The Study Material is strictly based on NCERT Latest Exam Solved Paper is included The Concepts are explained in depth Chapters are compiled with Previous Years Questions Answers to Questions included with Explanations Presence of accurate Figures throughout Five sets of Mock Tests are also included at the end This title focuses on an all inclusive preparations providing the aspirants to learn revise test and gauge their progress against the examination level The Book contains the following units Unit I Physical Chemistry I Unit II Physical Chemistry II Unit III Organic Chemistry I Unit IV Organic Chemistry II Unit V Inorganic Chemistry I Unit VI Inorganic Chemistry II *A Stoichiometry Unit* David Callaghan, 2004 **Inorganic Chemistry** Tina Overton, Jonathan Rourke, Fraser A. Armstrong, 2018 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Seventh Edition is the ideal course companion for the duration of a student s degree The authors have drawn upon their extensive teaching and research experience to update this text the seventh edition retains the much praised clarity of style and layout from previous editions while offering an enhanced section on expanding our horizons The latest innovative applications of green chemistry have been added to clearly illustrate the real world significance of the subject This edition also sees a greater used of learning features including substantial updates to the problem solving questions additional self tests and walk through explanations which enable students to check their understanding of key concepts and develop problem solving skills Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resources Inorganic Chemistry Seventh Edition is accompanied by a range of online resources For registered adopters of the text DT Figures marginal structures and tables of data ready to download DT Test bank For students DT Answers to self tests and exercises from the book DT Tables for group theory DT Web links DT Links to interactive structures and other resources on www.chemtube3d.com **Stoichiometry Unit Project** Luann Marie Decker, 1998 *A Crash Course in AIEEE Chemistry 2011* , **Objective NCERT Based Chapterwise Topicwise Solutions For 11th And 12th Class with Solved Papers (2005 -2023) with Notes for NEET-AIIMS Exam 2024 - Chemistry** Dr K.G. Ojha, 2023-07-14 Excel in Chemistry for NEET AIIMS Exam 2024 with this comprehensive guide featuring objective NCERT based solutions solved papers and notes for classes 11th and 12th Objective NCERT From Prabhat Exam is an unparallel book designed on the complete syllabus of 11th and 12th NCERT textbook It is the leading choice of Toppers

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