

## Unit 8 Worksheet 1: Mole relationships

For each of the problems below:

- Write the balanced chemical equation.
- Identify what is given (with units) and what you want to find (with units).
- Use coefficients from balanced equation to determine mole ratio.
- Show set up (organize it!).

- Hydrogen sulfide gas, which smells like rotten eggs, burns in air to produce sulfur dioxide and water. How many moles of oxygen gas would be needed to completely burn 8 moles of hydrogen sulfide?

|           |                                                                                                                                                                                                              |           |          |          |                        |
|-----------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----------|----------|----------|------------------------|
| Equation: | $\underline{\quad 2 \quad} \text{H}_2\text{S}_{(g)} + \underline{\quad 3 \quad} \text{O}_{2(g)} \rightarrow \underline{\quad 2 \quad} \text{SO}_{2(g)} + \underline{\quad 2 \quad} \text{H}_2\text{O}_{(g)}$ |           |          |          |                        |
| Before    | <u>2</u>                                                                                                                                                                                                     | ex        | <u>0</u> | <u>0</u> | 3.0 mol O <sub>2</sub> |
| Change    | <u>-2</u>                                                                                                                                                                                                    | <u>-3</u> | <u>2</u> | <u>2</u> |                        |
| After     | <u>0</u>                                                                                                                                                                                                     | ex        | <u>2</u> | <u>2</u> |                        |

- Propane, C<sub>3</sub>H<sub>8</sub>, burns in air to form carbon dioxide and water. If 12 moles of carbon dioxide are formed, how many moles of propane were burned?

Equation:  $2\text{AL} + 6\text{HCL} \rightarrow 2\text{ALCL}_3 + 3\text{H}_2$

|        |      |      |      |      |             |
|--------|------|------|------|------|-------------|
| Before | 0.5  | ex   | 0    | 0    | 1.50mol HCL |
| Change | -0.5 | -1.5 | +0.5 | +0.5 |             |
| After  | 0    | ex   | 0.5  | 0.75 |             |

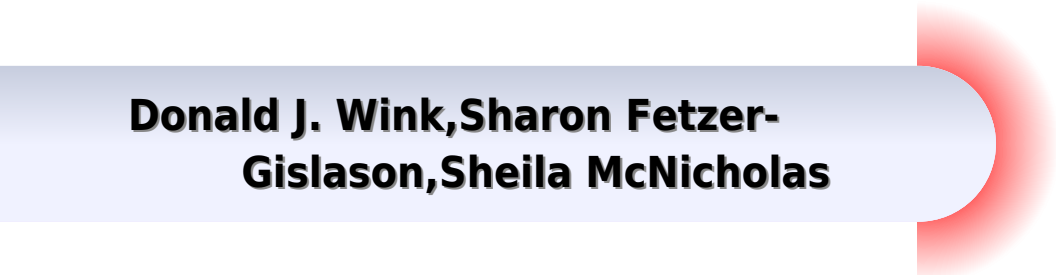
- Ammonia, NH<sub>3</sub>, for fertilizer is made by causing hydrogen and nitrogen to react at high temperature and pressure. How many moles of ammonia can be made from 0.15 moles of nitrogen gas?

Equation:  $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$

|        |       |       |     |           |
|--------|-------|-------|-----|-----------|
| Before | ex    | 1.25  | 0   | 2.50 mole |
| Change | -3.75 | -1.25 | +25 |           |
| After  | ex    | 0     | 2.5 |           |

# Unit 7 1 Mole Relationships Answer Key

**Donald J. Wink, Sharon Fetzer-  
Gislason, Sheila McNicholas**



## Unit 7 1 Mole Relationships Answer Key:

**Basic Concepts of Chemistry** Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance

*An Introduction to Chemistry* Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

*The Practice of Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink\\_POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink_POCPowerPoint_Final.ppt) 832KB

**Fundamentals of Physics, Volume 2** David Halliday, Robert Resnick, Jearl Walker, 2021-10-19 Renowned for its interactive focus on conceptual understanding its superlative problem solving instruction and emphasis on reasoning skills the Fundamentals of Physics Volume 2 12th Edition is an industry leading resource in physics teaching With expansive insightful and accessible treatments of a wide variety of subjects including photons matter waves diffraction and relativity the book is an invaluable reference for physics educators and students In the second volume of this two volume set the authors discuss subjects including Coulomb's Law Gauss Law and Maxwell's Equations

Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds

compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

**Prentice Hall Chemistry**, 2000

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**Practice Makes Perfect Chemistry** Marian DeWane, Heather Hattori, 2011-05-10 Don't be confused by chemistry Master this science with practice practice practice Practice Makes Perfect chemistry is a comprehensive guide and workbook that covers all the basics of chemistry that you need to understand this subject Each chapter focuses on one major topic with thorough explanations and many illustrative examples so you can learn at your own pace and really absorb the information You get to apply your knowledge and practice what you've learned through a variety of exercises with an answer key for instant feedback Offering a winning formula for getting a handle on science right away Practice Makes Perfect chemistry is your ultimate resource for building a solid understanding of chemistry fundamentals

**Chemistry Textbook for College and University USA** Ibrahim Sikder, 2023-06-04

**Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

**Toxic Cyanobacteria in Water** Ingrid Chorus, Martin Welker, 2021-03-07 Cyanobacterial toxins are among the hazardous substances most widely

found in water They occur naturally but concentrations hazardous to human health are usually due to human activity Therefore to protect human health managing lakes reservoirs and rivers to prevent cyanobacterial blooms is critical This second edition of Toxic Cyanobacteria in Water presents the current state of knowledge on the occurrence of cyanobacteria and cyanotoxins as well as their impacts on health through water related exposure pathways chiefly drinking water and recreational activity It provides scientific and technical background information to support hazard identification assessment and prioritisation of the risks posed by cyanotoxins and it outlines approaches for their management at each step of the water use system It sets out key practical considerations for developing management strategies implementing efficient measures and designing monitoring programmes This enables stakeholders to evaluate whether there is a health risk from toxic cyanobacteria and to mitigate it with appropriate measures This book is intended for those working on toxic cyanobacteria with a specific focus on public health protection It intends to empower professionals from different disciplines to communicate and cooperate for sustainable management of toxic cyanobacteria including public health workers ecologists academics and catchment and waterbody managers Ingrid Chorus headed the department for Drinking Water and Swimming Pool Hygiene at the German Environment Agency Martin Welker is a limnologist and microbiologist currently with bioMérieux in Lyon France

*Biological Science* Jon Scott, Anne Goodenough, Gus Cameron, Dawn Hawkins, Martin Luck, Jenny Koenig, Alison Snape, Despo Papachristodoulou, Kay Yeoman, Mark Goodwin, 2022 A fresh approach to biology centred on a clear narrative active learning and confidence with quantitative concepts and scientific enquiry Spanning the breadth of biological science and designed for flexible learning it will give you a deeper understanding of the key concepts and an appreciation of biology as a dynamic experimental science

**Holt Chemistry** Ralph Thomas Myers, 2004

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Introduction to Contextual Maths in Chemistry Fiona Dickinson, Andrew McKinley, 2020-10-23 For any student who has ever struggled with a mathematical understanding of chemistry this book is for you We include insights from real students which identify common problem areas and provide the prompts that helped them to overcome these

*Environmental Health Perspectives*, 1993

eBook: General, Organic and Biological Chemistry 2e SMITH, 2012-02-16

eBook General Organic and Biological Chemistry 2e

CRASH COURSE JEE(MAIN) / AIEEE - PHYSICS V&S EDITORIAL BOARD, 2015-01-09 This book is meant to be a quick refresher for JEE MAIN AIEEE aspirants With the aim and scope of providing a comprehensive study package for aspirants of JEE MAIN AIEEE this crash course focuses less on theory and

more on concepts formulae and tips This is supported by plenty of practice problems based on the latest formats structure and syllabus of JEE MAIN AIEEE This is further supplemented by a CD given along with this study kit with fully solved 2012 JEE MAIN AIEEE question paper Salient features A Based on the latest pattern and syllabus of JEE MAIN AIEEE A Solved examples practice problems in each chapter A Previous years question papers fully solved A Less theory and more concepts formulae and tips A Practice CD with fully solved JEE MAIN AIEEE 2012 question paper A Plenty of problems for practice A Comprehensive holistic revision of the complete syllabus of JEE MAIN AIEEE A In depth analysis of the recent trends of JEE MAIN AIEEE A A quick and efficient study kit for JEE MAIN AIEEE aspirants A Facilitates self study A Low priced handy book for quick and efficient revision      **Cumulated Index Medicus** ,2000      **Princeton Review AP Chemistry Premium Prep, 25th Edition** The Princeton Review,2023-08-01 EVERYTHING YOU NEED TO HELP SCORE A PERFECT 5 WITH THE BEST PRACTICE ON THE MARKET Equip yourself to ace the AP Chemistry Exam with this comprehensive study guide including 7 full length practice tests the MOST full length tests on the market thorough content reviews targeted strategies for every section and access to online extras Techniques That Actually Work Tried and true strategies to help you avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need for a High Score Fully aligned with the latest College Board standards for AP Chemistry Comprehensive content review for all test topics Engaging activities to help you critically assess your progress Access to study plans a handy list of key equations helpful pre college information and more via your online Student Tools Premium Practice for AP Excellence 7 full length practice tests 5 in the book 2 online with detailed answer explanations Practice drills at the end of each content review chapter Review of important laboratory procedures and equipment

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