

Unit 7 Chemistry - Nail Lab

Purpose *After you have completed the lab, think about what was the overall point for performing the lab. What were we trying to accomplish?*

Procedure

Day 1

1. Mass a clean & labeled _____ mL beaker.
2. Put between _____ g of copper (II) chloride in the beaker. Record the mass of the beaker and the copper (II) chloride.
3. Add _____ mL of distilled water to the beaker. Stir to dissolve the solid.
4. Obtain the mass of _____ iron nails together to ± 0.01 g.
5. Place the nails in the copper (II) chloride solution. Observe the reaction; record your observations. Place the labeled beaker in the place designated by your teacher.

Day 2

6. Remove the nails. Rinse or scrape the _____ (copper metal) from the nails into your labeled beaker. Place the nails in a separate labeled small beaker. Note the appearance of the nails in your observations. Place this beaker in the drying oven.
7. _____ the solution from the 250 mL beaker. Rinse the precipitate with about _____ mL of distilled water. Try to lose as little of the solid copper as you can when you decant. After a 2nd rinse with distilled water, rinse the copper with _____. Rinse one last time with distilled water. Then place the labeled beaker in the drying oven.
8. Mass the dry nails, and then discard the nails.

Day 3

9. Mass the beaker + dry copper. Discard the copper in the place designated by your teacher. Wash your beaker and let dry.

Data:

Mass 250 mL beaker	
Mass 250 mL beaker + copper (II) chloride	
Mass nails before reaction	
Mass nails after reaction	

Unit Nail Lab Chemistry Conclusion Answers

LP Steffe



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Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry The Popular Science News and Boston Journal of Chemistry ,1883 *Laboratory Manual [in] Engineering Physics ...* Seville Chapman,1947 *Scientific American* ,1917 **Bulletin of the Atomic Scientists** ,1972-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world Chemistry David W. Oxtoby,Wade A. Freeman,Toby F. Block,1998 Textbook of chemistry concepts Backpacker ,2007-09 Backpacker brings the outdoors straight to the reader s doorstep inspiring and enabling them to go more places and enjoy nature more often The authority on active adventure Backpacker is the world s first GPS enabled magazine and the only magazine whose editors personally test the hiking trails camping gear and survival tips they publish Backpacker s Editors Choice Awards an industry honor recognizing design feature and product innovation has become the gold standard against which all other outdoor industry awards are measured *The Chemical Trade Journal and Chemical Engineer* G Kelville Davis,1932

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