# Unit 3- thermochemistry

Study of Energy Changes that happen with matter changes State Changes first law of thermodynamia S-q. Sublimation total amount of energy in the S-E metting universe is constant 1-q. boiling exo thermic 9-5' doposition System - Products 9-1. Condensation Surroundings to all matter appromy/teleasing emengy System Types universe

= System + Surroundings

YSOUTHON TUPE Exchanges.

exothermic process

energy is produced / released Creation

Chdothermic Process

energy is absorbed (product)

## **Unit 1test Thermochemistry**

**Kaplan Test Prep** 

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