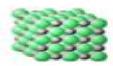
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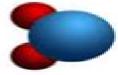
Activity #1- Introduction to Covalent Bonding

Open Chemical Bonding http://www.visionlearning.com/en/library/Chemistry/1/Chemical-Bonding/55 Scroll down to the heading "Covalent Bonding."

1. As opposed to ________bonding in which a complete transfer of electrons occurs, _______bonding occurs when two or more elements _______electrons. Covalent bonding occurs because the atoms in the compound have a similar tendency for electrons (generally to _______electrons). This most commonly occurs when two _______bond together. Because both of the nonmetals will want to ______electrons in an effort to ______their valence shells.

- 2. How many valence electrons are in one atom of hydrogen?
- 3.How many valence electrons does hydrogen need to have a full first shell?
- 4. How does the hydrogen atom "pick up" another electron?
- 5.What compound does hydrogen form?
- 6.How do hydrogen atoms make a covalent bond?
- 7. Label the substances below as "ionic" or "covalent"





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