

Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: a procedure for collecting, testing, and communicating information
2. Define hypothesis: a suggested answer to the question, an educated guess
3. Define experiment: a scientific test so see what will happen
4. Define dependent variable: the variable that you do NOT control. (on a graph is usually on the Y axis)
5. Define independent variable: the variable that I control. (on a graph is usually on the X axis)
6. Define control: the part of the experiment where everything is left the same.
7. Define observation: information collected with any of your 5 senses
8. Define density: the ratio of mass per unit volume ($D = m/v$)
9. Define accuracy: the measure of how close your data is to the ACTUAL or true value.
10. Define precision: a measure of how close your data (measurements) are to each other.
11. Why do scientists use SI units of measurement? So that measurements can be understood internationally (around the world).
12. List (in the correct order) the steps of the scientific method.

- | | |
|------------------------|-----------------------------------|
| 1. make an observation | 4. test hypothesis (experiment) |
| 2. ask a question | 5. analyze data/form a conclusion |
| 3. form a hypothesis | 6. develop a theory |

13. What mnemonic device is used to remember the metric system?

K **H** **Da** **B** **d** **c** **m**
g, m, L

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	K	H	Da	g, m, L	d	c	m
Meaning in powers of 10	10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}

Convert the following:

- | | |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL | 21. 25 cm to <u>0.25</u> m |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg | 23. 15 g to <u>0.015</u> kg |
| 19. 0.35 kg to <u>350</u> g | 24. 3.452 kL to <u>3452</u> L |
| 20. 300 mL to <u>0.300</u> L | 25. 47 m to <u>47000</u> mm |

Scientific Notation:

Convert the following to scientific notation:

- | | | | |
|------------|--------------------------------------|------------|--|
| 26) 0.005 | <u>5×10^{-3}</u> | 30) 0.025 | <u>2.5×10^{-2}</u> |
| 27) 5.050 | <u>5.05×10^1</u> | 31) 0.0025 | <u>2.5×10^{-3}</u> |
| 28) 0.0008 | <u>8×10^{-4}</u> | 32) 500 | <u>5×10^2</u> |
| 29) 1,000 | <u>1×10^3</u> | 33) 5,000 | <u>5×10^3</u> |

Unit 8 Study Guide Chemistry Answer Key

DP Hallahan



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Table of Contents Unit 8 Study Guide Chemisty Answer Key

1. Understanding the eBook Unit 8 Study Guide Chemisty Answer Key
 - The Rise of Digital Reading Unit 8 Study Guide Chemisty Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 8 Study Guide Chemisty Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit 8 Study Guide Chemisty Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 8 Study Guide Chemisty Answer Key
 - Personalized Recommendations
 - Unit 8 Study Guide Chemisty Answer Key User Reviews and Ratings

- Unit 8 Study Guide Chemistry Answer Key and Bestseller Lists
- 5. Accessing Unit 8 Study Guide Chemistry Answer Key Free and Paid eBooks
 - Unit 8 Study Guide Chemistry Answer Key Public Domain eBooks
 - Unit 8 Study Guide Chemistry Answer Key eBook Subscription Services
 - Unit 8 Study Guide Chemistry Answer Key Budget-Friendly Options
- 6. Navigating Unit 8 Study Guide Chemistry Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit 8 Study Guide Chemistry Answer Key Compatibility with Devices
 - Unit 8 Study Guide Chemistry Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 8 Study Guide Chemistry Answer Key
 - Highlighting and Note-Taking Unit 8 Study Guide Chemistry Answer Key
 - Interactive Elements Unit 8 Study Guide Chemistry Answer Key
- 8. Staying Engaged with Unit 8 Study Guide Chemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 8 Study Guide Chemistry Answer Key
- 9. Balancing eBooks and Physical Books Unit 8 Study Guide Chemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 8 Study Guide Chemistry Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 8 Study Guide Chemistry Answer Key
 - Setting Reading Goals Unit 8 Study Guide Chemistry Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 8 Study Guide Chemistry Answer Key
 - Fact-Checking eBook Content of Unit 8 Study Guide Chemistry Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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