Chemistry 12 Review Sheet on Unit 1 -Reaction Kinetics

Looking at the expressions for reaction rate on page 1 SW, write similar expressions with which
you could express rates for the following reactions. (Hint: look at what happens to reactants and
products.) Recall that solid or liquids can lose or gain mass, gases can lose or gain volume and
aqueous solutions can increase or decrease in concentration. ("a" is done as an example.)

a)
$$Mg_{(s)} + 2HCl_{(aq)} \Rightarrow H_{2(g)} + MgCl_{2(aq)}$$

reaction rate = $\frac{mass}{mass} \frac{mass}{mass} \frac{mass}{mass}$

c)
$$C_{(s)} + O_{2(g)} \rightarrow CO_{2(g)}$$

Unit 12 Review Sheet Answers Chemistry

Tobias Bleicker

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