

Summative Assessment: Nuclear Chemistry – Radioactivity & Nuclear Energy (Traditional Organization)

Matching: Match the description in column A with the correct term in column B. Write the letter in the blank provided. Each term matches with only one description, so be sure to choose the best description for each term.

Column A

- ___ **L** ___ 1. Used in the medical field to help provide information about how the body functions
- ___ **G** ___ 2. Helium nucleus
- ___ **C** ___ 3. Self-sustaining process
- ___ **M** ___ 4. Instrument used for measuring radioactivity levels
- ___ **D** ___ 5. Undergoes spontaneous nuclear decay
- ___ **I** ___ 6. High-energy photon
- ___ **A** ___ 7. Amount of radioactive material necessary for fission to occur
- ___ **N** ___ 8. Time it takes for half of a sample of radioactive nuclei to decay
- ___ **E** ___ 9. Splitting one heavy nucleus into two nuclei with smaller mass numbers
- ___ **H** ___ 10. Electron
- ___ **J** ___ 11. Particle with the same mass as an electron but a positive charge
- ___ **B** ___ 12. Subatomic particles found in the nucleus
- ___ **F** ___ 13. Combining two light nuclei to form a heavier nucleus
- ___ **K** ___ 14. The change of one element into another

Column B

- A. Critical Mass
- B. Nucleons
- C. Chain reaction
- D. Radioactive
- E. Fission
- F. Fusion
- G. Alpha particle
- H. Beta particle
- I. Gamma ray
- J. Positron
- K. Nuclear transmutation
- L. Radiotracers
- M. Geiger Counters
- N. Half-life

Multiple Choice - Choose the one best answer and write its letter in the left margin.

- ___ **C** ___ 15. Identify the missing particle: $^{239}_{94}\text{Pu} \rightarrow \text{_____} + ^4_2\alpha$
- A. $^{235}_{92}\text{U}$
 B. $^{239}_{92}\text{Th}$
C. $^{235}_{92}\text{U}$
 D. $^{239}_{96}\text{Am}$

- ___ **B** ___ 16. Identify the missing particle: $^{238}_{88}\text{Ra} \rightarrow \text{_____} + ^4_2\alpha$

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