

Unit 8 – Stoichiometry Worksheet 1: Mole relationships

For each of the problems below, use the five steps discussed in class. Answer on a separate piece of paper.

1. Write the balanced chemical equation
2. Identify what is given (with units) and what you want to find (with units) in a BCA table
3. Use the mole ratio from the balanced equation to calculate remaining values in the Table (dimensional analysis style). **Show your work beneath the BCA table.**
4. Complete the BCA Table using calculation results
5. Find the desired quantity in your work. Write and box in that answer with unit, substance.

1. Hydrogen sulfide gas, which smells like rotten eggs, burns in air to produce sulfur dioxide and water. How many moles of oxygen gas would be needed to completely burn 8.0 moles of hydrogen sulfide? (The BCA table is done as an example. Make/copy your own on your own piece of paper).

Equation: $\text{___ H}_{2}\text{S}_{(g)} + \text{___ O}_{2(g)} \rightarrow \text{___ SO}_{2(g)} + \text{___ H}_2\text{O}_{(g)}$

Before:

FILL IN THE VALUES ON YOUR OWN PAPER

Change

(SHOW WORK UNDER THE TABLE)

After

2. Propane, C_3H_8 , burns in air to form carbon dioxide and water. If 12 moles of carbon dioxide are formed, how many moles of propane were burned?
3. Ammonia, NH_3 , for fertilizer is produced by causing hydrogen and nitrogen to react at high temperature and pressure. How many moles of ammonia can be made from 0.15 moles of nitrogen gas?
4. The poison gas phosgene, COCl_2 , reacts with water in the lungs to form hydrochloric acid and carbon dioxide. How many moles of hydrochloric acid would be formed by 0.835 moles of phosgene?
5. Nitric acid, HNO_3 , is used in the production of trinitrotoluene, (TNT), an explosive. The equation (not balanced) is



- a) How many moles of nitric acid are required to react with 4.5 moles of toluene? (make your own BCA table)
 - b) How many moles of TNT should be produced?
6. Iron metal and oxygen combine to form magnetite, the magnetic oxide of iron, Fe_3O_4 .
 - a) How many moles of iron can be converted to magnetite by 8.80 moles of pure oxygen? (make your own BCA table)
 - b) How many moles of the magnetic iron oxide would be produced?

Unit 8 Stoichiometry Answers

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Unit 8 Stoichiometry Answers:

Solutions Manual to Accompany Inorganic Chemistry Alen Hadzovic, 2018 As you master each chapter in Inorganic Chemistry having detailed solutions handy allows you to confirm your answers and develop your ability to think through the problem solving process **Resources in Education** ,1984-10 **Laboratory Manual to Accompany Chemistry, [by] Stanley R. Radel, Marjorie H. Navidi** Arthur D. Baker, 1990 *Ebook: Chemistry* Julia Burdge, 2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text Solving Problems in Chemistry Rod O'Connor, Charles Mickey, Alton Hassell, 1977 A Textbook of Biotechnology For Class XI Dr. R.C. Dubey, Multiple choice questions with their answers are also incorporated to help students preparing for competitive examinations **Problems and Solutions for General Chemistry and College Chemistry, Sixth Editions by Nebergall, Holtzclaw, and Robinson** Frederick K. Ault, 1980 **Objective NCERT Based Chapterwise Topicwise Solutions For 11th And 12th Class with Solved Papers (2005 -2023) with Notes for NEET-AIIMS Exam 2024 - Chemistry** Dr K.G. Ojha, 2023-07-14 Excel in Chemistry for NEET AIIMS Exam 2024 with this comprehensive guide featuring objective NCERT based solutions solved papers and notes for classes 11th and 12th Objective NCERT From Prabhat Exam is an unparallel book designed on the complete syllabus of 11th and 12th NCERT textbook It is the leading choice of Toppers and the pinnacle for NEET exam along with NCERT This book is a must for NEET BOARDS CUET as it has questions extracted from each and every line of the NCERT textbook Extra Notes are added from experts to make it more understandable Chapter wise NCERT notes for quick yet thorough impactful revisions Tabular texts Illustrative diagrams in HD pages for understanding NCERT Based Topic wise MCQs from each of NCERT to get firm grip on concepts NCERT Exemplar Problem MCQs to develop a strong base go in depth Assertion Reason Case Based Questions HOTS to cover all question typologies Exam Archive including Previous years NEET other PMT exam s questions Practice Papers Model Test Papers to put final practice touch to your preparation 5 Mock Test to Make you an experienced player Answer keys hints and explanations are also added in the book for micro level understanding Competition Science Vision ,1999-12 Competition Science Vision monthly magazine is published by Pratiyogita Darpan Group in India and is one of the best Science monthly magazines available for medical entrance examination students in India Well qualified professionals of Physics Chemistry Zoology and Botany make contributions to this magazine and craft it with focus on providing complete and to the point study material for aspiring candidates The magazine covers General Knowledge Science and Technology news Interviews of toppers of examinations study material of Physics Chemistry Zoology and Botany with model papers reasoning test questions

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Dr Manish Rannjan IAS the author of NCERT Objective Textbook Chemistry is an esteemed educator and accomplished civil servant With his extensive knowledge and experience Dr Rannjan aims to simplify the study of chemistry for students and provide them with a comprehensive resource that aligns with the NCERT curriculum His book is a valuable tool for students preparing for various competitive exams enabling them to develop a solid foundation in chemistry and achieve academic success

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spain won their second consecutive title by defeating croatia 22 20 in the final