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## Acids/Bases & pH Worksheet

#### PROPERTIES OF ACIDS AND BASES

- 1. Acids taste \_\_\_\_\_\_ and bases taste \_\_\_\_\_\_, bases also feel \_\_\_\_\_\_
- 2. Acids turn litmus paper \_\_\_\_\_\_, bases turn litmus paper \_\_\_\_\_ and phenolphthalein \_\_\_\_
- Circle: Weak acids are <u>strong / weak</u> electrolytes. A weak electrolyte is a <u>poor / good</u> conductor of electricity.

#### REACTIONS OF ACIDS AND BASES

4. The definitions of a Brønsted-Lowry acid is: \_\_\_\_\_

Identify the Brønsted-Lowry Acid & Base, and also the conjugate acid and conjugate base in each reaction:

- 5.  $NH_3(g) + H_2O(0) \rightleftharpoons NH_4^+(sq) + OH^-(sq)$
- 6.  $HClO_{3(aq)} + H_2O_{(1)} \rightleftharpoons H_3O^*_{(aq)} + ClO_3^-_{(aq)}$
- 7. HF(aq) +  $HSO_3^-(aq)$   $\rightleftarrows$   $F^-(aq)$  +  $H_2SO_3(aq)$
- 8. The product of an Arrhenius acid and base neutralization are a/an \_\_\_\_\_\_ and \_\_\_\_\_\_

Complete and balance the following reactions:

- 9. \_\_\_\_ HBr + \_\_\_ NaOH → \_\_\_\_\_ + \_\_\_\_\_
- 10. H<sub>2</sub>SO<sub>4</sub> + KOH → +
- 11. \_\_\_\_ HCl + \_\_\_ Ca(OH)<sub>2</sub> → \_\_\_\_\_ + \_\_\_\_
- 12. \_\_\_\_ Fe(OH)<sub>3</sub> + \_\_\_\_ H<sub>2</sub>SO<sub>4</sub> → \_\_\_\_\_ + \_\_\_\_

#### NAMING ACIDS & BASES:

- 13. What is the correct formula for nitrous acid?
- 14. What is the correct formula for hydrobromic acid? \_\_\_\_\_\_
- 15. What is the correct name for H<sub>3</sub>PO<sub>3</sub>?
- 16. What is the correct name for HNO<sub>3</sub>?
- 17. What is the correct formula for bromic acid? \_\_\_\_\_\_
- 18. What is the correct name for H<sub>2</sub>SO<sub>4</sub>?
- 19. What is the formula for aluminum hydroxide? \_\_\_\_\_\_
- 20. What is the correct name for Sr(OH)<sub>2</sub>? \_\_\_\_\_\_
- 21. What is the correct name for Cu(OH)<sub>2</sub>? \_\_\_\_\_\_
- 22. What is the formula for iron (III) hydroxide? \_\_\_\_\_

# Worksheet 10 1 Acid Base And Ph

**DP Hallahan** 

#### Worksheet 10 1 Acid Base And Ph:

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