

Osmosis and Tonicity

Define osmosis. _____

In which direction does water move across membranes, up or down the concentration gradient? _____

Define these 3 terms:

- a. isotonic- _____
- b. hypertonic _____
- c. hypotonic _____

Use arrows to show the direction of water movement into or out of each cell. Color and label the cell in an isotonic environment light blue, the hypotonic environment yellow, and the hypertonic environment light green.



Match the description or picture with the osmotic condition:

- | | | |
|---------------|-------|--|
| A. Isotonic | _____ | solution with a lower solute concentration |
| | _____ | solution in which the solute concentration is the same |
| B. Hypertonic | _____ | condition plant cells require |
| | _____ | condition that animal cells require |
| C. Hypotonic | _____ | red blood cell bursts (cytolysis) |
| | _____ | plant cell loses turgor pressure (Plasmolysis) |
| | _____ | solution with a higher solute concentration |
| | _____ | plant cell with good turgor pressure |
| | _____ | solution with a high water concentration |

Tonicity And Osmosis Worksheet Key

Sir Tarun Rupani

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Educart ICSE Class 10 One-shot Question Bank 2026 Biology (strictly for 2025-26 boards) Sir Tarun Rupani, 2025-07-12

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Isotonic, hypotonic, and hypertonic solutions Rumi Michael Leigh, 2022-08-21

Isotonic Hypotonic and Hypertonic Solutions Things You Should Know Questions and Answers introduces the principles of fluid balance osmosis and electrolyte solutions in a question and answer format The book explains what solutes and solvents are how concentration works and how terms such as hyper hypo and iso describe different solution types Readers will learn about cations anions ions osmosis cell membranes and the difference between intracellular and extracellular fluid It also describes plasmolysis cytolysis and osmotic lysis along with the effects of hypertonic hypotonic and isotonic solutions on cells Clinical uses are outlined such as giving hypertonic solutions in cases of cerebral edema or hyponatremia and the risks of hypotonic solutions causing cell swelling or lysis The book highlights how isotonic solutions maintain equilibrium prevent fluid shifts and increase extracellular fluid volume without disturbing the balance inside the cell This book will be useful for nursing students medical students and healthcare workers who want to understand isotonic hypotonic and hypertonic solutions their effects on cells and their role in patient care

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