

7. A chemical system that has no output or input of matter but does have an input/output of energy is known as:
- A closed system
 - An open system
 - An isolated system
 - None of the above

8. Which statement concerning the diagram below is true?
- ΔH is positive
 - The system is endothermic
 - Energy is released to the surroundings
 - The enthalpy of the products is greater than the enthalpy of the reactants



9. In the reaction $2 \text{NO} (\text{g}) + \text{O}_2 (\text{g}) \rightarrow 2 \text{NO}_2 (\text{g})$, if the concentration of nitrogen dioxide changes from 0.45 mol/L to 1.00 mol/L in 2 minutes, what is the overall rate of production of nitrogen dioxide in the system?
- 3.64 mol/(L•min)
 - 0.275 mol/(L•min)
 - 0.137 mol/(L•min)
 - 12.6 mol/(L•min)
 - 0.333 mol/(L•min)

10. The activated complex...
- is an unstable molecule
 - has the maximum potential energy possible
 - may revert to reactants
 - all of the above
11. A catalyst works by:
- changing the products that are formed in the reaction
 - decreasing the activation energy of the reaction
 - decreasing the enthalpy change of the reaction
 - changing the enthalpy of the reactants
 - increasing the activation energy of the reaction

12. Which of the following could be a rate law for a fourth order reaction?
- $r = k[X]^4[Y]^1$
 - $r = k[X]^4[Y]^4$
 - $r = k[X]^2[Y]^2$
 - $r = k[X]^1[Y]^2$

13. What property could help indicate the reaction rate for the following reaction?



$$\Delta H = -44.51 \text{ kJ/mol}$$

- Change in pressure
- Change in color
- Change in temperature
- Change in mass

Thermodynamics And Reaction Rates Review Answer

Nikhilesh Mukherjee



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Catalog of National Bureau of Standards Publications, 1966-1976: Citations and abstracts United States. National Bureau of Standards. Technical Information and Publications Division, 1978

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Chemistry of Polysaccharides Gennady Zaikov, 2005-04-18 This monograph summarizes scientific achievements in the field of polysaccharide chemistry performed in the last decade For the first time the particular features of polysaccharide structure physical chemical properties and biochemical transformations in their interrelations are considered as well as the questions of polysaccharide modification along the whole hierarchical chain including starch mono di oligo and polysaccharides chitin chitozan and cellulose Polysaccharide interactions with diverse solvents and low molecular macrocyclic and high molecular substances in solutions are scrutinized Several aspects of mechanical chemical modification of polysaccharides in solid state and in gel are observed Modern concepts of enzyme degradation of polysaccharides and technologies of natural polymers processing are analyzed The book is intended for wide range of readers students post graduates engineers and scientists engaged in polymer chemistry organic and physical chemistry

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Supercritical Fluid Engineering Science Erdogan Kiran, Joan F. Brennecke, American Institute of Chemical Engineers.

Meeting,1993 Developed from papers presented at the Symposium on Supercritical Fluids held at the American Institute of Chemical Engineers Annual Meeting in Los Angeles November 1991 this volume reports on recent developments and reflects the diversity and expanding scope of applications of supercritical fluids The first part is devoted to phase behavior thermodynamics and transport properties the second part to recent research on molecular interactions modeling and computer simulations and the final part to more specific applications including polymers pharmaceuticals coal and petroleum products environmental remediation and chromatography Annotation copyright by Book News Inc Portland OR

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