

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

Answer: B

16. According to kinetic theory of gases, the absolute zero temperature is attained when

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

Answer: C

17. Kinetic theory of gases assumes that the collisions between the molecules are

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

Answer: A

18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to

- A. E/3
- B. E/2
- C. 2E/3
- D. 3E/2
- E. 3E

Answer: C

19. Kinetic energy of the molecules in terms of absolute temperature T is proportional to

- A. T
- B. T^2
- C. $T^{1/2}$
- D. $T^{3/2}$
- E. $T^{5/2}$

Answer: A

20. Supercooled vapour behaves

- A. exactly as gas
- B. as liquid
- C. as solid
- D. as solid-liquid vapour
- E. supercooled as a gas

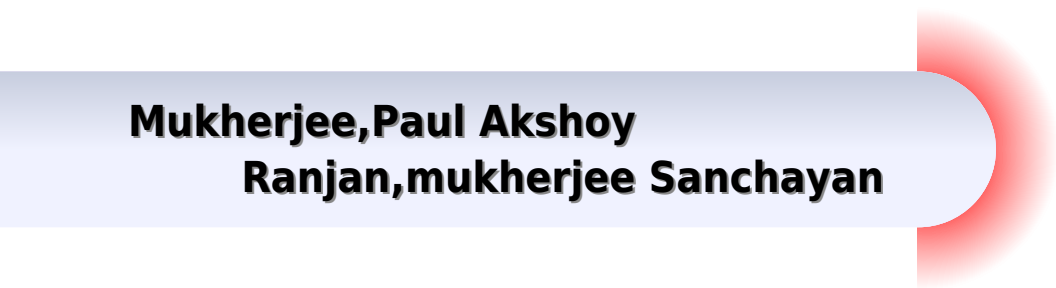
Answer: A

21. Absolute zero pressure will occur

- A. at absolute zero
- B. at the centre of the earth

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Mukherjee,Paul Akshoy
Ranjan,mukherjee Sanchayan



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Thermodynamics S. C. Gupta, 2005-09

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