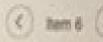
# Chapter 6. Thermochemistry Assessment Adaptive Follow-Up

# For Practice 6.2 - Enhanced - with Feedback



# III Review | Constants | Periodic Table

To determine whether a shiny gold-colored rock is actually gold, a chemistry student decides to measure its heat capacity. She first weighs the rock and finds it has a mass of 4.4 g. She then finds that upon absorption of 55.6 J of heat, the temperature of the rock rises from 25.1 C to 64.1 C.

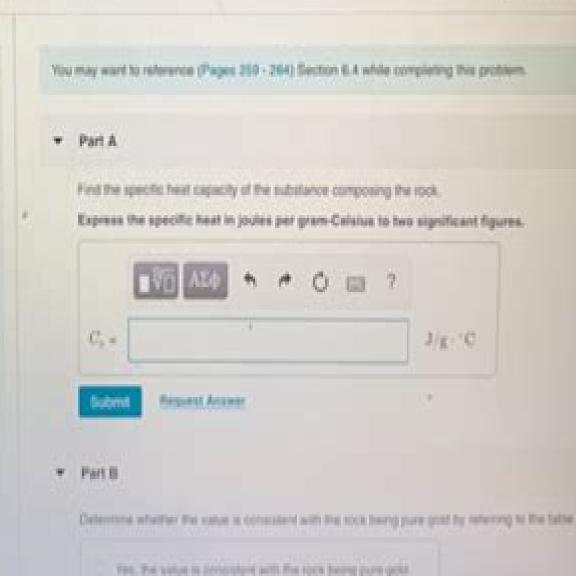
Substance	Specific capacity (J/(g)	heat y.C. C))*
Elements		

0.526 Last Cold 0.126 Shar 0.235 **Display** 0.345 bear. 0.440 **FAMILIANIA** 9.303 Compounds **Ethansi** 2.62 Water 4.18

6.7%

Materials

Glass (Pyrox)



Not, this value is inconsistent with the sock being pure gold.

# **Thermochemistry Assessment Answers**

**RM Cervero** 

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