

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It varies in the periodic table.
- _____ 2. It does not vary in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are strong oxidizing agents.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 1. There are two representative elements of elements.
- _____ 2. There are two representative elements of the elements in the periodic table.
- _____ 3. Group 1 elements are highly reactive metals.
- _____ 4. Group 17 elements are the most reactive metals.
- _____ 5. Group 1 elements are highly reactive metals.
- _____ 6. Group 17 elements are highly reactive metals.
- _____ 7. Group 1 elements are highly reactive metals.
- _____ 8. Group 17 elements are highly reactive metals.
- _____ 9. Group 1 elements are highly reactive metals.
- _____ 10. Group 17 elements are highly reactive metals.
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- _____ 12. Group 17 elements are highly reactive metals.
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- _____ 99. Group 1 elements are highly reactive metals.
- _____ 100. Group 17 elements are highly reactive metals.

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