

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	1
_____	2
_____	3
_____	4
_____	5
_____	6
_____	7

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 16. It increases in the periodic table.
- _____ 17. It decreases in the periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are shiny and conduct electricity.
- _____ 20. Group 17 elements.

Column B

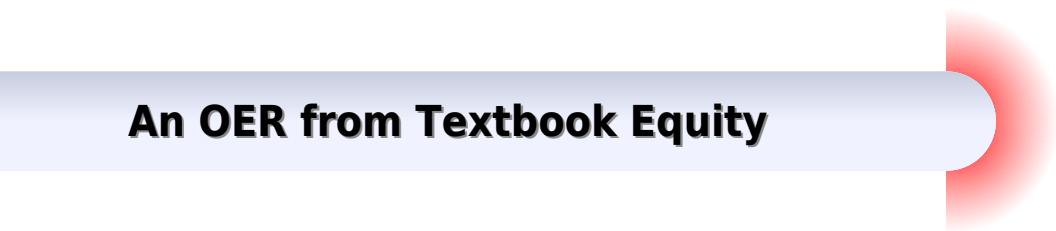
- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in bold; change the underlined word or phrase as needed to make it true.

- _____ 21. There are two representative elements of phosphorus.
- _____ 22. More than three families of the elements in the periodic table are metalloids.
- _____ 23. Group 1A elements except for hydrogen are located on the left side.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are inert nonreactive elements known as noble gases.
- _____ 27. Alkaloids have properties of both metals and more reactive metals.

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