

chapter (3)

Equilibrium System

(Seen)

(unSeen)

- Stationary System apparently but dynamic on reality

Saturated water vapour Pressure : max pressure at certain temperature

Condensation $\leftrightarrow$ evaporation

water vapour pressure at certain temp

Types of chemical reaction

- ① complete (irreversible) : reaction occur in one direction due to releasing gas or ppt of Salts,

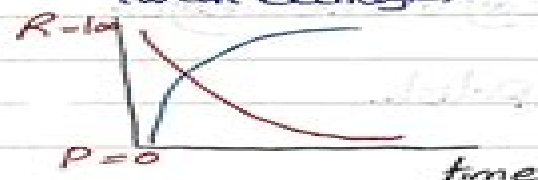


closed vessel
reversible



$\hookrightarrow$ non ionized in Soluble

Ionic equation
(Weak electrolyte)



concentration of reactant $\downarrow$

" " " " Products $\uparrow$

- * There is complete reaction in Neutralization bet. Strong acid and Strong base



Study Guide For Gentic Equilibrium

**David E. Sadava, Gordon H.
Orians, Craig Heller, William K. Purves**



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