

UNIT 4: CHEMICAL REACTIONS STUDY GUIDE

Multiple choice:

___**C**___1. Chemical reactions _____.

- a. simply change the form or state of the reactants.
- b. create new atoms.
- c. always produce new substances.
- d. create more matter than what they started with.

___**B**___2. Which of the following is the correct skeleton equation for the combination of solid zinc with liquid bromine to produce solid zinc bromide?

- a. $\text{Zn}_{2(s)} + \text{Br}_{(l)} \rightarrow \text{ZnBr}_{2(s)}$
- b. $\text{Zn}_{(s)} + \text{Br}_{2(l)} \rightarrow \text{ZnBr}_{2(s)}$
- c. $\text{Zn}_{(s)} + \text{Br}_{2(aq)} \rightarrow \text{ZnBr}_{2(s)}$
- d. $\text{ZnBr}_{2(s)} \rightarrow \text{Zn}_{(s)} + \text{Br}_{2(l)}$

___**A**___3. A substance that is needed to speed up a reaction but is not a product or a reactant is known as a _____.

- a. catalyst
- b. coefficient
- c. spectator ion
- d. steroid

___**D**___4. In a combustion reaction, one of the reactants is always _____.

- a. a liquid
- b. water
- c. a metal
- d. oxygen

___**B**___5. Which of the following statements is true about the decomposition of a simple binary compound?

- a. The products cannot be predicted.
- b. The reactant is a single substance.
- c. The reactant must always be a molecular substance.
- d. The product is a single substance.

___**C**___6. The reaction $2\text{P}_2\text{O}_5 \rightarrow 4\text{P} + 5\text{O}_2$ is an example of which type of reaction?

- a. double replacement
- b. single replacement
- c. decomposition
- d. combustion
- e. synthesis

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Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

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Schore, 2002-08-02 **Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic

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Chemistry, Student Study Guide James E. Brady,Fred Senese,2008-01-28

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