

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
39
55
87
101
119

atomic mass	19.
atomic number	39.
electron configuration	55.
chemical name	87.
chemical symbol	119.

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It exists in the periodic table.
- _____ 19. It has no bar periodic table
- _____ 20. Group 1 elements
- _____ 21. Elements that are strong oxidizing ability
- _____ 22. Group 8 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most of the elements of the periodic table are metals.
- _____ 25. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 29. Alkaloids have properties of both metals and nonmetals.

Study Guide For Content Mastery Stoichiometric

Neil D. Jespersen, Alison Hyslop



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