

Name: _____ Date: _____

MASS TO MOLE STOICHIOMETRY

Answer the questions. Write answers on new paper.

1. In photosynthesis, water reacts with carbon dioxide to give oxygen and glucose ($\text{C}_6\text{H}_{12}\text{O}_6$). Write and balance the chemical equation. How many moles of CO_2 are required to make 120.0g of glucose?



a. What is the mass of potassium nitrate that is produced when 2.04 moles of potassium phosphate react?

b. If 580g of aluminum phosphate are formed, how many moles of aluminum nitrate reacted?



a. If 150g of N_2O_4 was produced, how many moles of O_2 were required?

b. If 4.0×10^3 moles of oxygen reacted, how many grams of N_2 were needed?

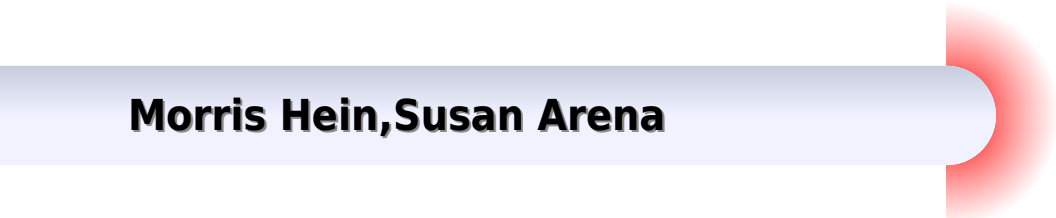


a. If you have 5.50 mol of CaC_2 , how much C_2H_2 do you get?

b. How many moles of water are needed when 65.0g of CaC_2 have reacted?

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Morris Hein, Susan Arena



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