

# 10.1 Periodic Trends



## 10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

### Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|       |
|-------|
| _____ |
| _____ |
| _____ |
| _____ |
| _____ |

|                        |       |
|------------------------|-------|
| Atomic Mass            | _____ |
| Atomic Number          | _____ |
| Electron Configuration | _____ |
| Chemical Name          | _____ |
| Chemical Symbol        | _____ |

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 18. It exists in the periodic table.
- \_\_\_\_\_ 19. It has no line periodic table.
- \_\_\_\_\_ 20. Group 1 elements.
- \_\_\_\_\_ 21. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 22. Group 17 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of elements.
- \_\_\_\_\_ 24. Most elements have higher of the elements in the periodic table are metals.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 1A elements are highly reactive elements known as alkalis.
- \_\_\_\_\_ 28. Group 1A elements are very reactive elements known as alkalis.
- \_\_\_\_\_ 29. Alkaloids have properties of both metals and non-metals.

# Study Guide For Content Mastery Ch 5

**Richard Bailey**



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