

Thermochemistry Problems

1) Consider the following reaction:



- a) Is the reaction exothermic or endothermic?
- b) Calculate the amount of heat released if 2.4 g of Mg(s) reacts.
- c) How many grams of MgO are produced during an enthalpy change of 96.0 kJ?
- d) How many kilojoules of heat are absorbed when 7.50 g of MgO(s) are decomposed into Mg(s) and O₂(g) at constant pressure?

2) Choose endothermic and exothermic reactions below:

- I) $\text{HgO} \rightarrow \text{Hg} + \text{O}_2$
- II) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
- III) $\text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{NO}_2 + \text{O}_2$
- IV) $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
- V) $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$

3) Choose endothermic and exothermic processes:

- A) Condensation
- B) Evaporation
- C) Boiling
- D) Freezing
- E) Sublimation

4) Determine the ΔH for each of the following reactions.



Thermochemistry Final Test

David Kirk



Thermochemistry Final Test:

United States Air Force Academy United States Air Force Academy, 1972 **SAT Subject Test Chemistry** Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron's SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory ONLINE PRACTICE TESTS Students who purchase this book or package will also get access to two additional full length online SAT Chemistry subject tests with all questions answered and explained **SAT Subject Test: Chemistry with Online Tests** Joseph A. Mascetta, Mark Kernion, 2018-09-01 Always study with the most up to date prep Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product *Annual Catalog - United States Air Force Academy* United States Air Force Academy, 1971 *Energy Research Abstracts* , 1995

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