



Chapter 5- Thermochemistry

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Enthalpy

[Enthalpy diagram](#)

[Bond enthalpy](#)

[Dissolution reaction](#)

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Enthalpy

	Exothermic	Endothermic
energy	energy is released from the system → temp of reaction vessel becomes warmer	energy is absorbed into the system → temp of reaction vessel becomes colder
sign of ΔH	negative	positive

$$\Delta H = H_{\text{products}} - H_{\text{reactants}} = x[\text{kJ/mol}]$$

Enthalpy diagram

Thermochemistry Chapter 5

F Rizvi



Thermochemistry Chapter 5:

The Enigmatic Realm of **Thermochemistry Chapter 5**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Thermochemistry Chapter 5** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those who partake in its reading experience.

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