

Thermochemistry Test Review: Multiple Choice Problems

- What is the correct order, from highest to lowest molar heat value, for the three types of energy change?
 - Chemical, phase, nuclear
 - Phase, chemical, nuclear
 - Nuclear, phase, chemical
 - Nuclear, chemical, phase
- Which metal requires the most energy to raise 1.00 g of it by 1.00°C? (Refer to Chemistry Data Booklet)
 - Aluminum
 - Copper
 - Lead
 - Nickel
- What is the specific heat capacity of a substance if 2.41×10^4 J are needed to change the temperature of 105.0 g of it from 25.0°C to 250.0°C?
 - $1.02 \times 10^{-4} \text{ J/g}^\circ\text{C}$
 - $9.18 \times 10^{-4} \text{ J/g}^\circ\text{C}$
 - $0.918 \text{ J/g}^\circ\text{C}$
 - $1.02 \text{ J/g}^\circ\text{C}$
- Which statement about enthalpy is true?
 - Heat is given off to the surroundings in endothermic reactions.
 - Some substances have a negative specific heat capacity.
 - Specific heat capacity is the same for all liquids.
 - The sign of ΔH is always negative in exothermic reactions.
- What happens to the value of ΔH for a thermochemical reaction if the reaction is reversed?
 - ΔH has the same numerical value, and the sign changes.
 - ΔH has the same numerical value, and the sign remains the same.
 - ΔH is the reciprocal of the original value, and the sign changes.
 - ΔH is the reciprocal of the original value, and the sign remains the same.
- Which is an exothermic process?
 - Ice melting
 - Water boiling
 - Water evaporating
 - Water vapour condensing
- Which statement is true for the combustion of ethanol?

$$\text{C}_2\text{H}_5\text{OH}(\text{l}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 3\text{H}_2\text{O}(\text{l}) \quad \Delta H = -1370 \text{ kJ}$$
 - The enthalpy change would be the same if gaseous water were produced.
 - The potential energy of the products is less than the potential energy of the reactants.
 - The products of the reaction occupy a larger volume than the reactants.
 - The reaction is endothermic.
- What is the energy required to evaporate two moles of liquid water given the following equations?

$$\begin{aligned} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g}) &\rightarrow 2\text{H}_2\text{O}(\text{g}) & \Delta H = -483.6 \text{ kJ} \\ 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g}) &\rightarrow 2\text{H}_2\text{O}(\text{g}) & \Delta H = -571.6 \text{ kJ} \end{aligned}$$
 - 44.0 kJ
 - 88.0 kJ
 - 527.6 kJ
 - 1055.2 kJ
- Which statement correctly describes an endothermic chemical reaction?
 - The products have higher potential energy than the reactants, and the ΔH is negative.
 - The products have higher potential energy than the reactants, and the ΔH is positive.
 - The products have lower potential energy than the reactants, and the ΔH is negative.
 - The products have lower potential energy than the reactants, and the ΔH is positive.
- Given the thermochemical equation,

$$2\text{NO}_2(\text{g}) \rightarrow 2\text{NO}(\text{g}) + \text{O}_2(\text{g}) \quad \Delta H = -114 \text{ kJ}$$
 what is the ΔH for the reaction $\text{NO}(\text{g}) + \frac{1}{2}\text{O}_2(\text{g}) \rightarrow \text{NO}_2(\text{g})$?
 - 114 kJ
 - 57 kJ
 - +57 kJ
 - +114 kJ
- Using the data below, what is the order of changes that occur when ethanol is heated from -25.0°C to 85.0°C?

Boiling point of ethanol: 78.3°C
Melting point of ethanol: -117.3°C

 - Phase change, temperature change
 - Phase change, temperature change, phase change
 - Temperature change, phase change
 - Temperature change, phase change, temperature change
- Which statement best describes the formation of methanol, $\text{CH}_3\text{OH}(\text{l})$, from the elements? (Refer to Data Booklet, page 5)
 - It is endothermic and heat is absorbed.
 - It is exothermic and heat is released.
 - It is exothermic and heat is absorbed.
 - It is exothermic and heat is released.
- Which process results in the greatest endothermic change for 10.0 g of H_2O ?
 - Condensation
 - Melting
 - Solidification
 - Vaporization
- The following decomposition reaction may occur in an air bag:

$$2\text{NaN}_3(\text{s}) \rightarrow 3\text{N}_2(\text{g}) + 2\text{Na}(\text{s}) \quad \Delta H = -43.5 \text{ kJ}$$
 What is the heat of formation, ΔH_f° , for NaN_3 ?
 - 43.5 kJ
 - 21.8 kJ
 - 21.8 kJ
 - 43.5 kJ
- A small sample released 2.0×10^{11} kJ of energy while undergoing a change. What type of change most likely occurred?
 - Chemical
 - molecular
 - nuclear
 - physical

Thermochemistry Multiple Choice Test

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Thermochemistry Multiple Choice Test:

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web have the same problem with mine the relay is behind the panel on the passenger side down where the heater is there are 3 relays in the same spot the magnetic relay was the top one on mine i replaced it but clutch still did not engage

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web jun 15 2004 check the fuse for the parking lights fuses are behind the fuse panel on the passenger side of the middle console but i dont remember a relay for tail lights sorry anyway check all your fuses make sure none of them are blown

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web in this article we will show the locations of the electronic control units a description of the fuses and relays of the toyota celica 230 with box diagrams and photo examples of their execution highlight the cigarette lighter fuse

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web nov 21 2009 one is the main relay 1 in the fuse block in the engine bay and the other one is the circuit opening relay behind the drivers side kick panel you actually have to take the panel off to get to it it will be the green relay on the bottom right corner there is a 10mm bolt right above it that the mount goes to

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web dec 15 2009 this sounds stupid but is there a diagram somewhere that shows what each of the relays in the fuseboxes do i am specifically looking for the headlight relay we recently had the low beams both stop working in our 2003 and i figured it was either the relay or the switch

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web aug 4 2016 passenger s side kick panel toyota celica fuse box passenger s side kick panel fuses type a fuses type b fuses type c warning terminal and harness assignments for individual connectors will vary depending on vehicle equipment level model and market

ac clutch relay location toyota celica forum - Apr 30 2022

web mar 16 2015 ok so i figured out that the relay is in a box next to the radiator putting power to that relay engages the ac

clutch which is just below it clutch works relay works but when i press the ac button the clutch and fans do not engage
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web nov 9 2021 here you will find the locations and photos of the fuse blocks separately we note the elements responsible for the cigarette lighter and fuel pump in the engine compartment

help circuit opening relay toyota celica forum - Aug 03 2022

web jun 15 2015 1 5 of 5 posts shadowgriever 2 jun 14 2015 i checked in the factory service manual and it shows the circuit opening relay is under the front of the center console i would like to mention this as well since i am not sure how you have checked for power to your fuel pump

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web fuel pump relay deceleration sensor the main relay of the electric drive of the roof folding audio amplifier start inhibit switch relay block no 4 mirror control relay mode relay ex hi fuse box it is located at the bottom of the dashboard on the driver s side behind the protective cover