

Temperature

- A substance of mass _____ J/kg absorbed the particles that make up the substance.
- Absorbed energy is the product of mass _____ in the sample.
- $Q = m \times c \times \Delta T$

Heat

- Small _____ amount of _____ energy that flows between two _____.
- Absorbed into _____.
- $Q = m \times c \times \Delta T$

Specific energy is _____.

Potential energy is _____.

- Potential energy is the energy that an object has because of its _____.

Specific Heat Capacity

- Amount of heat _____ required to raise the _____ of 1 kg of substance by 1°C.
- $Q = m \times c \times \Delta T$

$$Q = m \times c \times \Delta T$$

$$Q = m \times c \times \Delta T$$

Change in temperature

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